

Lab 2 - Scientific Measurement

OBJECTIVES:

- Make quantitative and qualitative observations.
- Understand the importance of the Metric System (International System of Units) and how it is used in unit conversions.
- Identify and learn how to use standard laboratory tools used for measuring volume and mass.
- Utilize scientific method to investigate accuracy and precision of measuring tools.
- Calculate mean and standard deviation.
- Understand how polarity of various substances affect other substances in a mixture.
- Utilize the scientific method to investigate the different properties of water.

MATERIALS:

- Two decimal place laboratory balance
- 1 mL Transfer Pipet
- 10 mL Serological pipette and pump
- 10 mL Graduated cylinder
- 250 mL Beaker with deionized water
- 50 mL Beaker
- 2 Paperclips
- Petri dish
- 2 Glass slides
- Petroleum ether
- Vegetable oil
- Table Salt

SAFETY:

Understand and adhere to safe laboratory practices when performing any activity in the classroom or lab. Pay attention to any demonstrations of the protocol for the correct use of the instruments and materials necessary to complete the activities. Use personal protective equipment such as safety glasses or goggles, gloves, and aprons when appropriate. In this lab activity you will be using glassware, which may break if dropped, so please handle equipment with care.

BACKGROUND:

As a biology student, you will often be asked to make observations. These observations will be of either a qualitative or quantitative nature. A qualitative observation is a descriptive representation of a characteristic and is not numerical. Quantitative observations are numerical. Although qualitative observations can be valuable, quantitative observations are easier to evaluate and are preferred in laboratory science. For example, if you were studying rainfall in Florida and wanted to determine which city had the most rainfall last year, you could provide a quantitative description, e.g. 52 inches.

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Table of Contents **Bsc1010l Manual**

1. Understanding the eBook Bsc1010l Manual
 - The Rise of Digital Reading Bsc1010l Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Bsc1010l Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Bsc1010l Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Bsc1010l Manual
 - Personalized Recommendations
 - Bsc1010l Manual User Reviews and Ratings
 - Bsc1010l Manual and Bestseller Lists

5. Accessing Bsc1010l Manual Free and Paid eBooks
 - Bsc1010l Manual Public Domain eBooks
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 - ePub, PDF, MOBI, and More
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 - Bsc1010l Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Bsc1010l Manual
 - Highlighting and Note-Taking Bsc1010l Manual
 - Interactive Elements Bsc1010l Manual
8. Staying Engaged with Bsc1010l Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Bsc1010l Manual
9. Balancing eBooks and Physical Books Bsc1010l Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Bsc1010l Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Bsc1010l Manual
 - Setting Reading Goals Bsc1010l Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Bsc1010l Manual
 - Fact-Checking eBook Content of Bsc1010l Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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