

Chapter 1

Photosynthesis

INTRODUCTION

Central Concept: The first law of thermodynamics states that energy cannot be created or destroyed, only the form can be changed. Living organisms do not “make” energy; they convert it from an environmental source (e.g., sunlight, food molecules) instead. The second law states that these energy conversions are not 100% efficient, i.e., there is a decrease in available energy with each conversion.

The balanced equation for photosynthesis is as follows:



This means that inorganic molecules (carbon dioxide and water) can be assembled into glucose if the energy of sunlight is supplied. Oxygen and water are also produced. Photosynthesis is one of the most important metabolisms in the biological world. As a result of this reaction, green plants accomplish three major things: energy of sunlight is converted into chemical energy, inorganic molecules are converted into organic molecules, and oxygen is released as an end product. Let's look at each in more detail.

1. The energy of sunlight is trapped by a plant pigment called chlorophyll during photosynthesis; this trapped energy is then formed into the chemical bonds that hold organic molecules together. When these bonds are broken at a future time, their energy can be released to power the growth, reproduction, and other energy-requiring processes of the plant. Since many animals (including humans) eat plants, the energy in these organic molecules is used to power the metabolism of animals as well.
2. Photosynthesis is one way of converting inorganic molecules into organic molecules. Inorganic molecules (CO_2 and H_2O) are used to produce an organic molecule called PGAL (= **p**hosphog**ly**cer**al**dehyde). PGAL is a three-carbon molecule that the plant can (with the proper enzymes, inorganic molecules, and organic molecules) convert into amino acids, fatty acids, and glucose. Only green plants, algae, and a few species of bacteria can turn inorganic molecules into organic ones.
3. Free oxygen gas (O_2) is released into the environment as a product of photosynthesis. Plants and animals require this oxygen for aerobic cellular respiration, which is the way the bonds in glucose are broken down to release energy.

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N Colangelo



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Increasing the NADPH supply for whole-cell biotransformation and development of a novel biosensor Solvej Siedler,2013 **In-situ Biotransformation of Carbon Tetrachloride Under Anoxic Conditions** Lewis Semprini,1991

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Manual of Microbiology, Biochemistry and Molecular Biology J. Saxena,M. Baunthiyal,I. Ravi,2015-05-01 Though many practical books are available in the market but this Laboratory Manual of Microbiology Biochemistry and Molecular Biology is an unique combination of protocols that covers maximum about 80% of the practicals of various Indian universities for UG and PG courses in Bioscience Biotechnology Microbiology Biochemistry and Biochemical Engineering **A Laboratory**

Manual in Elementary Biology Emanuel Roth Boyer,1894 Biotransformation and Metabolite Elucidation of Xenobiotics Ala F. Nassar,2011-03-23 The goal of this book is to improve the readers knowledge of metabolite elucidation in drug metabolism by exposing them to in depth coverage of the biotransformation of xenobiotics strategies for identifying and characterizing metabolites FDA guidelines and case studies on how to improve the decision making process in structural modification of drug candidates to reduce toxicity The book consists of 8 chapters it first provides an introduction on biotransformation of xenobiotics and then presents modern approaches and strategies for dealing with metabolite characterization using tools such as LC MS H D exchange stable isotopes LC MS NMR and radiolabeled compounds Also strategies for dealing with reactive intermediates in drug discovery and development are presented as well as case studies on improving the decision making process in the structural modification of drug candidates The last chapter discusses the regulatory perspectives of safety testing of drug metabolites and why how and when to test their safety **Biology**

Laboratory Manual Sylvia S. Mader,2000-07 Mader includes revised coverage of animal behaviour and ecology as well as a

wealth of new focus boxes which highlight topics of high interest and relate biology to everyday life This text is linked to a web site offering extended chapter outlines *Laboratory Manual of Biological Chemistry* Otto Folin,1919

Biotransformation and Fate of Chemicals in the Aquatic Environment Alan W. Maki,Kenneth L. Dickson,John Cairns,1980 Environmental exposure concentration of chemicals background Evaluation of methods Bioavailability and the material balance approach to estimating environmental concentrations Environmental extrapolation of biotransformation data Quantitative expression of biotransformation Fate of chemicals in the aquatic environment case studies Workshop summary and conclusions **Exploring Genetics and Developmental Biology Using**

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