

Addis Ababa Science and Technology University
College of Mechanical and Electrical Engineering
Department of Electrical and Computer Engineering

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ECEg3102

Lecture I

Feedback Amplifiers

Applied Electrical Concepts

Debasish Dutta



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Plants A. Kayode Coker,2014-11-29 The fourth edition of Ludwig s Applied Process Design for Chemical and Petrochemical Plants Volume Three is a core reference for chemical plant and process engineers and provides an unrivalled reference on methods process fundamentals and supporting design data New to this edition are expanded chapters on heat transfer plus additional chapters focused on the design of shell and tube heat exchangers double pipe heat exchangers and air coolers Heat tracer requirements for pipelines and heat loss from insulated pipelines are covered in this new edition along with batch heating and cooling of process fluids process integration and industrial reactors The book also looks at the troubleshooting of process equipment and corrosion and metallurgy Assists engineers in rapidly analyzing problems and finding effective design methods and mechanical specifications Definitive guide to the selection and design of various equipment types including heat exchanger sizing and compressor sizing with established design codes Batch heating and cooling of process fluids supported by Excel programs

Meat Science and Applications Y. H. Hui,Wai-Kit Nip,Robert Rogers,2001-07-27 Meat Science and Applications compiles the most recent science technology and applications of meat products by products and meat processing It details worker safety waste management slaughtering carcass evaluation meat safety and animal handling issues from an international perspective Essential concepts are illustrated with practical examples and helpful diagrams

DIY Off-Grid Solar Power for Everyone Debasish Dutta,2024-01-24 DIY Off Grid Solar Power for Everyone is a comprehensive guide crafted for anyone looking to embrace the world of solar energy This book simplifies the journey into off grid solar power making it accessible to both beginners and experienced enthusiasts

Vacuum and Ultravacuum Igor Bello,2017-11-02 Vacuum technology has enormous impact on human life in many aspects and fields such as metallurgy material development and production food and electronic industry microelectronics device fabrication physics materials science space science engineering chemistry technology of low temperature pharmaceutical industry and biology All decorative coatings used in jewelries and various daily products including shiny decorative papers the surface finish of watches and light fixtures are made using vacuum technological processes Vacuum analytical techniques and vacuum technologies are pillars of the technological processes material synthesis deposition and material analyses all of which are used in the development of novel materials increasing the value of industrial products controlling the technological processes and ensuring the high product quality Based on physical models and calculated examples the book provides a deeper look inside the vacuum physics and technology

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise

calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectivesto correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objekte can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound's normal boiling point refers to its boiling point at a pressure of 760 mm Hg

experiment 11 vapour pressure proctech 2ce3 lab - Jul 07 2022

the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I aim to figure out the boiling point of a mystery substance in order to identify it chemically this experiment requires the handling of a Bunsen burner boiling

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there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0.5ml of material

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web junior has struggled and often failed to find acceptance there his entire life he feels a responsibility to prove himself to rowdy and the other native americans that see him as a traitor at the same time junior tries to make a new name for himself at the all white reardan high school

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decides to transfer to the school in reardan because of a conversation with mr p a white teacher whose nose he has broken by throwing a textbook across the room mr p forgives junior for breaking his nose but asks for

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