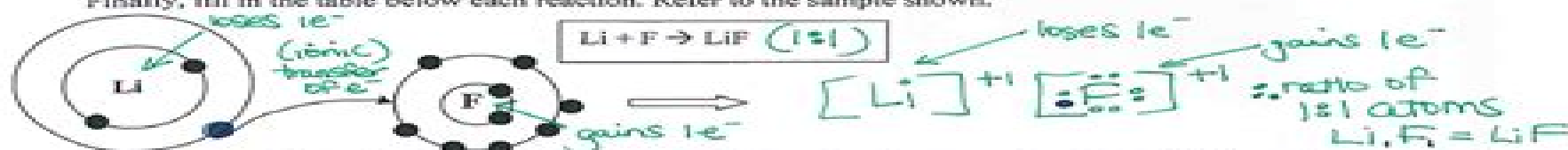


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Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers 1ve ⁻ to F	Li ⁺
F	7 ve	accepts 1ve ⁻ from Li	F ⁻

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$ 	Li	1 ve	donate 1e ⁻	Li ⁺
	Cl	1 ve	accepts 1e ⁻	Cl ⁻
2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$ 	Ca	2 ve	donates 2e ⁻	Ca ⁺²
	O	6 ve	accepts 2e ⁻	O ⁻²
3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$ 	Be	2 ve	donates 2e ⁻ to each F	Be ⁺²
	F	7 ve	each F accepts 1ve each	F ⁻
4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$ 	Mg	2 ve	donates 2e ⁻ to S	Mg ⁺²
	S	6 ve	accepts 2e ⁻	S ⁻²
5) $\text{K} + \text{F} \Rightarrow \text{KF}$ 	K	1 ve	donates 1ve ⁻	K ⁺¹
	F	7 ve	accepts 1ve ⁻	F ⁻¹

Bond Formation Study Guide Answers

**John Krenos, Joseph Potenza, Carl
Hoeger**



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Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

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Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

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scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place Helps readers learn to categorize analyze and solve organic chemistry problems at all levels of difficulty Hundreds of fully worked practice problems all with solutions Key concept summaries for every chapter reinforces core content from the companion book Study Guide and Solutions Manual Neil E. Schore,2002-08-02 **Organic Chemistry, 5e Student**

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groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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