

13. (4 pts) What is an ionic compound?
- A compound that will dissolve in water
 - Two nonmetal atoms bonded together
 - Two metal atoms bonded together
 - A metal and a nonmetal atom bonded together
 - None of the above
14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
16. (4 pts) How many total atoms are in this molecule: $C_6H_{12}OH$?
- 3
 - 4
 - 19
 - 20
 - None of the above
17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $C_5H_{10}Cl_x$, what must x be?
- 1
 - 2
 - 3
 - 4
 - None of the above
18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C_3H_6ClF $C_6H_5CH_3$ CS_2 $BaSO_4$
- C_3H_6ClF
 - $C_6H_5CH_3$
 - CS_2
 - $BaSO_4$
 - CS_2 and $BaSO_4$

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1. Understanding the eBook Apologia Chemistry Module 15 Study Guide
 - The Rise of Digital Reading Apologia Chemistry Module 15 Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Apologia Chemistry Module 15 Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Apologia Chemistry Module 15 Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Apologia Chemistry Module 15 Study Guide
 - Personalized Recommendations
 - Apologia Chemistry Module 15 Study Guide User Reviews and Ratings
 - Apologia Chemistry Module 15 Study Guide and Bestseller Lists
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- 7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Apologia Chemistry Module 15 Study Guide
 - Interactive Elements Apologia Chemistry Module 15 Study Guide
- 8. Staying Engaged with Apologia Chemistry Module 15 Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Apologia Chemistry Module 15 Study Guide
- 9. Balancing eBooks and Physical Books Apologia Chemistry Module 15 Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Apologia Chemistry Module 15 Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Apologia Chemistry Module 15 Study Guide
 - Setting Reading Goals Apologia Chemistry Module 15 Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Apologia Chemistry Module 15 Study Guide
 - Fact-Checking eBook Content of Apologia Chemistry Module 15 Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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