PHYSICAL PHARMACEUTICS

QUESTIONS:

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- a) Mass. b) Volume. c) Temperature. d)Pressure.
- When a large amount of solute dissolves in a given amount of solvent it forms:
 - a) Delute solution.
 - b) Saturated solution.
 - c) Aqueous solution.
 - d) Concentrated Solution.
- Solution which can hold no more solute is called:
 - a) Delute solution.
 - b) Saturated solution.
 - d) Aqueous solution.
 - d) Concentrated Solution.
- 4. Which is not a colligative property:
 - a) Roult's law.
 - b) Depression in freezing point.
 - d) Osmotic pressure.
 - d) Melangpoint.
- Azeotropic mixtures are:
 - a) Constant temperature boiling mixtures.
 - Those which boils at different temperatures.
 - c) Mixture of two solids.
 - d) None of above.
- The pH of pharmaceutical buffer system can be calculated by:
 - a) pH partition theory.
 - b) Nopes Whitney law.
 - d) Henderson Hasselbaltch equation.
 - d) Michaelia Menten equation.
- 7. Unit of surface tension is:
 - a) Dyney meter
 - b) Dyne/cm
 - d) Calam
 - d) None of the above.
- S. Two solutions are said to be isotonic if they exert same......
 - a) Viscosity.

- c) O smotic pressure.
- b) Surface tension.
- d) None of the above.
- 9. Dissolution is affected by:
 - a) Surface area.
 - b) Viscosity.
 - c) Temperature.
 - d) All of the above.

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G Psacharopoulos

Basic Physical Pharmacy Questions With Answer:

The book delves into Basic Physical Pharmacy Questions With Answer. Basic Physical Pharmacy Questions With Answer is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Basic Physical Pharmacy Questions With Answer, encompassing both the fundamentals and more intricate discussions.

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