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web describe how a mole is used in chemistry relate a mole to common counting units convert moles to number of representative particles and number of representative particles to moles vocabulary mole avogadro s number section 11 1 measuring matter if you were buying a bouquet of roses for a special occasion you probably

15 the mole is a measure of amount chemistry libretexts - Jul 04 2023

web it provides a specific measure of the number of atoms or molecules in a bulk sample of matter a mole is defined as the amount of substance containing the same number of discrete entities such as atoms molecules and ions as the number of

atoms in a sample of pure $12\ c$ weighing exactly $12\ g$