# **Biological Molecules**

#### Organic and Inorganic compounds

- Organic compounds are based on \_\_\_\_\_ and are found in living things. There are a number of
  exceptions including hydrogen carbonate (HCO<sub>1</sub> ), carbon dioxide (CO<sub>2</sub> )and Carbon monoxide (CO).
- compounds are by default are all the molecules other than those in the category above.

#### Common organic molecules

Monosaccharide sugars. These are the \_\_\_\_\_\_\_ from which larger \_\_\_\_\_\_ molecules are constructed. Molecules like glucose and fructose are metabolically active molecules usually stored in an inactive, insoluble polysaccharide form.

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- <u>Glucose</u>: C<sub>0</sub>H<sub>12</sub>O<sub>6</sub> is a \_\_\_\_\_ sugar (six carbons) most commonly found in this ring structure.
- Glucose will be known to most students as a product of or the substrate molecule for
- Glucose is also found in a polymer form of starch, glycogen and cellulose.
- •All bonds are
- Glucose is a reducing sugar and will give positive (Brick red) precipitate in a test.
- ·Glucose is metabolically active compound
- Glucose is \_\_\_\_\_ and has osmotic effects when in solution.

Ribose: (5 carbon sugar).

(hydrogen bonding) when in solution.

A modified version of ribose, deoxyribose is perhaps best known for its role in Deoxyribonucleic acid or \_\_\_\_\_ where it forms part of the sugar phosphate backbone. The chemical properties of deoxyribose are very \_\_\_\_ from the properties of ribulose. Both Ribous and Glucose will attract water molecules.

Amino Acids: There are \_\_\_\_\_common amino acids found in the protein structures of living things. Amino acids are monomers which combine to form the larger \_\_\_\_\_, in turn polypeptides combine to form \_\_\_\_. Proteins molecules are the basis of \_\_\_\_ and many cellular and extra cellular components.



Amino
NH2 — C — COOH
carboxylic
H acid

This model shows the structure of the general amino acid. If you build one in a molecular kit you will appreciate better the 3D structure. Each of the common amino acids has the same structure as the one shown except that the R group is different. Amino acids are soluble.

This is an alternative way to draw the general amino acid structure. This diagram illustrates the 'amino' group which is -NH<sub>2</sub>. There is also the acidic group -COOH which ionizes in solution to form an -COO and H'groups. This acid group is known as a carboxylic acid group.

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Keywords -

Management

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Hexiose:

Piolivimier.

Protose:

Covalent Carbon

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keywords to fill.

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**Albert Kaskel** 

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Improving the Experimental Skills of High School Biology Students by Introducing Laboratory Techniques of Molecular Biology Mary Margaret Fowler, 1989 

An Introduction to the Structure of Biological Molecules John Michael Barry, E. M. Barry, 1969

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