

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.	Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.

Unit 11: Chemical Equilibrium

Use the following information and complete the structure of Chemical Equilibrium.

Chemical Equilibrium	Chemical Equilibrium
Unit 11: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.	Unit 11: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.

Unit 12: Chemical Equilibrium

Use the following information and complete the structure of Chemical Equilibrium.

Chemical Equilibrium	Chemical Equilibrium
Unit 12: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.	Unit 12: Chemical Equilibrium Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. Indicate the effect of the pressure. 2. Indicate the effect of the volume. 3. Indicate the effect of the temperature. 4. The gas is at 1 atm. 5. The gas is at 2 atm.

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Name: _____

Unit 1: Chemical Equilibrium

Draw the best answer from those provided to each question.

Draw the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers.

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Convert the following into Scientific Notation.

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Chemical Equilibrium (Round your answers to 3 places after the decimal. Show work to show your work.)

a. Convert the pH value into [H⁺].

b. Convert the [H⁺] value into pH.

Unit 2: Chemical Equilibrium

Draw the best answer from those provided to each question.

Draw the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers.

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Convert the following into Scientific Notation.

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Chemical Equilibrium (Round your answers to 3 places after the decimal. Show work to show your work.)

a. Convert the pH value into [H⁺].

b. Convert the [H⁺] value into pH.

Unit 3: Chemical Equilibrium

Draw the best answer from those provided to each question.

Draw the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers.

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Convert the following into Scientific Notation.

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Chemical Equilibrium (Round your answers to 3 places after the decimal. Show work to show your work.)

a. Convert the pH value into [H⁺].

b. Convert the [H⁺] value into pH.

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Andrew Harvey



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Keep It Real With PBL, Secondary Jennifer Pieratt, 2019-12-19 Let's Get Real About PBL The book's companion website features an updated guide to help teachers integrate technology into PBL experiences for online and blended learning instruction Does project based learning PBL feel just out of reach in your secondary classroom Is project planning an overwhelming project in and of itself Dr Jennifer Pieratt a consultant and former teacher knows firsthand how challenging designing projects can be especially for secondary teachers with large caseloads and short class periods to engage in meaningful teaching and learning In this hands on interactive guide Pieratt supports secondary teachers through the iterative process of planning authentic project based learning experiences Using backward design she gives teachers ready to use strategies for identifying the best concepts to tackle in PBL experiences brainstorming realistic projects facilitating meaningful learning and creating formative and summative assessments The book is visually accessible in style and features reallife soundbites that tackle the challenges to implementing PBL Tips and resources to support the project planning process Planning forms to guide you through planning your projects Key terminology and acronyms in PBL Exercises to help you reflect and process throughout your project plans Master PBL planning with this clear efficient and easy to use guide to creating enriching experiences for your students *The United States Manual of Biography and History* James V.

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