

Table 18.2

Solubility Product Constants ( $K_{sp}$ ) at 25°C

| Salt              | $K_{sp}$                         | Salt                  | $K_{sp}$              | Salt                | $K_{sp}$              |
|-------------------|----------------------------------|-----------------------|-----------------------|---------------------|-----------------------|
| <b>Halides</b>    |                                  | <b>Sulfates</b>       |                       | <b>Hydroxides</b>   |                       |
| AgCl              | $1.8 \times 10^{-10}$            | PbSO <sub>4</sub>     | $6.3 \times 10^{-7}$  | Al(OH) <sub>3</sub> | $3.0 \times 10^{-34}$ |
| AgBr              | $5.0 \times 10^{-13}$            | BaSO <sub>4</sub>     | $1.1 \times 10^{-10}$ | Zn(OH) <sub>2</sub> | $3.0 \times 10^{-16}$ |
| AgI               | $8.3 \times 10^{-17}$            | CaSO <sub>4</sub>     | $2.4 \times 10^{-5}$  | Ca(OH) <sub>2</sub> | $6.5 \times 10^{-6}$  |
| PbCl <sub>2</sub> | $1.7 \times 10^{-5}$             | <b>Sulfides</b>       |                       | Mg(OH) <sub>2</sub> | $7.1 \times 10^{-12}$ |
| PbBr <sub>2</sub> | $2.1 \times 10^{-6}$             |                       |                       | Fe(OH) <sub>2</sub> | $7.9 \times 10^{-16}$ |
| PbI <sub>2</sub>  | $7.9 \times 10^{-9}$             |                       |                       | <b>Carbonates</b>   |                       |
| PbF <sub>2</sub>  | $3.6 \times 10^{-8}$             |                       |                       |                     |                       |
| CaF <sub>2</sub>  | $3.9 \times 10^{-11}$            |                       |                       |                     |                       |
| <b>Chromates</b>  |                                  |                       |                       |                     |                       |
|                   | PbCrO <sub>4</sub>               | $1.8 \times 10^{-14}$ |                       |                     |                       |
|                   | Ag <sub>2</sub> CrO <sub>4</sub> | $1.2 \times 10^{-12}$ |                       |                     |                       |
|                   |                                  | NiS                   | $4.0 \times 10^{-20}$ |                     |                       |
|                   |                                  | CuS                   | $8.0 \times 10^{-37}$ |                     |                       |
|                   |                                  | Ag <sub>2</sub> S     | $8.0 \times 10^{-51}$ |                     |                       |
|                   |                                  | ZnS                   | $3.0 \times 10^{-23}$ |                     |                       |
|                   |                                  | FeS                   | $8.0 \times 10^{-19}$ |                     |                       |
|                   |                                  | CdS                   | $1.0 \times 10^{-27}$ |                     |                       |
|                   |                                  | PbS                   | $3.0 \times 10^{-28}$ |                     |                       |

# A Solubility Product Constant Lab 17a Answers

**Katrin Zwirgmaier**



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