

## Physical Science: Midterm Study Guide Modules 1-7

### Module 1:

What is physical science?  
What is technology?  
What is the scientific method?  
Scientific Theory vs. Law  
Relationships shown on graphs (inverse, direct, none)

Quantitative observation  
Qualitative observation  
Hypothesis  
Variable  
Scientific Theory  
Inference

metric system units for length, mass, volume

### Module 2:

Solution  
Solute  
Solvent  
Solids  
Liquids  
Gases  
Density  
Diffusion

Differences between pure substances, compound, homogeneous mixtures, heterogeneous mixtures  
What is matter?  
Phases of matter: solid, liquid, gas, plasma phase-changes and their names (see pg 95)  
Properties of matter  
How can you tell if a physical vs. chemical change took place?

### Module 3:

Law of conservation of mass  
Law of constant composition  
Atomic number  
Mass number  
Isotopes  
Cations  
Anions/electrolytes

History of the atom: Democritus, Aristotle, Dalton's Atomic Theory, J.J. Thomson, Rutherford, Chadwick  
The atom - subatomic particles and their charges, atomic mass & number, how to figure out all the numbers given some information, isotopes and how to know how many neutrons there are given the name or how to name it given the number of protons  
Bohr's electron model with energy levels, table 2.2 pg 97  
Periodic table- how it's organized

# Biology Apologia Module 12 Study Guide

**Douglas Fraser, Maurice DiGiuseppe**



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**Comprehensive Dissertation Index** ,1984      *Books in Series* ,1985 Vols for 1980 issued in three parts Series Authors and Titles      **The British National Bibliography** Arthur James Wells,1994      Biology 12 Roger Prior,2010      *Biology 12* ,2012      **Biology 12 Study Guide** Susan Gerards,2011-12-29      **Biology 12** Stephen Gloury,2001      **The Key** ,2004  
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