Section Review 3.1

Part A Completion

- 1. accuracy 6. absolute value
- peecision 7, 100%
- 3. experimental value 8. scientific notation
- 4. error 9. known
- 5. accepted value 10. estimated

Part B True-False

11. ST 13. AT

Part C Matching

14. b 16. a 18. d 15. e 17. f 19. c

Part D Questions and Problems

- 20. a. 3 c. 2
 - b. 4
- 21. a. $4.35 \times 10^2 \, \mathrm{mL}$ e. 2.1×10^2
 - b. 8.4 cm³

Section Review 3.2

Part A Completion

- 1. metric
- seven
 Celsius
- 3. meter 8. kelvin
- 4. liter 9. joule/calorie 5. weight 10. calorie/joule

Part B True-False

11. NT 13. AT 12. NT 14. NT

Part C Matching

- 15. k 20. l 16. e 21. d
- 17. m 22. c
- 18. g 23. h 19. a 24. i

Part D Questions and Problems

- 28. Volume = $1.8 \text{ cm} \times 8.8 \text{ cm} \times 30.5 \text{ cm}$ = $4.8 \times 10^{2} \text{ cm}^{3}$
- 29, °C = K 273 = 20 273 = -253°C
- 30. a. h; 10²
- c. m 10⁻¹

25, 1

2566. (1)

27. b

6. gramm

- $b_{\rm s}$ k: 10^{3}
- $d_{\rm c} \approx 10^{-2}$

Section Review 3.3

Part A Completion

- 1. one punity:
- 6. steps
- 2. conversion factor
- 7. conversion factor

1.5 AT

- 3. remains the same
- 8. denominator
- dimensional analysis
- 10. unknown

9. umits

5. known

Part B True-False

- II. NT
- 13. AT
- 12. ST 14. NT

Part C Questions and Problems

16. a. $125 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 0.125 \text{ kg}$

$$= 1.25 \times 10^{-1} \, \mathrm{kg}$$

- **b.** $0.12 \text{ L} \times \frac{1000 \text{ mL}}{1 \text{ E}} = 1.2 \times 10^{2} \text{ mL}$
- 17. $\frac{1.0 \text{ jri}}{1500 \text{ cells}} \times \frac{2.54 \text{ cris}}{1 \text{ jri}} \times \frac{1 \text{ pri}}{100 \text{ ceris}} \times \frac{10^6}{1 \text{ jri}}$ = 17 μ m/cell
- 18, $\frac{186\,000\,\text{spdes}}{1\,\text{s}} \times \frac{1\,\text{s}}{10^6\,\mu\text{s}} \times \frac{1.61\,\text{km}}{1\,\text{gade}}$ = $\frac{2.99 \times 10^{-1}\,\text{km}}{1}$

Land

Section Review 3.4

Part A Completion

- 1. density 3. composition
- 2. intensive

Part B True-False

4. ST 5. ST

Part C Questions and Problems

6. $32.1 \text{ mL} \times \frac{1 \text{ cm}^3}{1 \text{ geV}} = 3.21 \text{ cm}^3$

Density =
$$\frac{\text{mass}}{\text{volume}} = \frac{127 \text{ g}}{32.1 \text{ cm}^3} = 3.96 \text{ g/cm}^3$$

7. $1.094. \times \frac{1000 \text{ cm}^3}{1 \text{ K}} = 1000 \text{ cm}^3$

$$1.58 \text{ kg} \times \frac{1000 \text{ g}}{\text{kg}} = 1580 \text{ g}$$

Density =
$$\frac{\text{mass}}{\text{volume}} = \frac{1580 \text{ g}}{1000 \text{ cm}^3} = 1.58 \text{ g/cm}^3$$

picture.

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JoAnn Zerwekh

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