Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 - Arrhenius Acids and Bases

| Common Name for Aqueous Solution | Chemical Formula | Found in | Tastes | Tums Litmus Paper | Conducts Electricity? | Acid or Base |
|-------------------------------------|------------------------|-------------------|--------|-------------------------|--------------------------|--------------------|
| Acetic acid | HC,H,O,(aq) | Vinegar | Sour | Red | Yes | Acid |
| Benzoic acid | HC,H,O,(49) | Food preservative | Sour | Red | Yes | Acid |
| Phosphoric acid | H,PO ₄ (aq) | Soda pop | Sour | Red | Yes | Acid |
| Hydrochloric acid | HCf(aq) | Stomach acid | Sour | Red | Yes | Acid |
| Citric acid | H,C,H,O,(49) | Citrus fruits | Sour | Red | Yes | Acid |
| Ascorbic acid | H,C,H,O,(aq) | Vitamin C | Sour | Red | Yes | Acid |
| Magnesium hydroxide | Mg(OH),(aq) | Milk of magnesia | Bitter | Blue | Yes | Base |
| Aluminum hydroxide | Al(OH),(aq) | Antacids | Bitter | Blue | Yes | Base |
| Barium hydroxide | Ba(OH),(aq) | Lubricants | POISON | Blue | Yes | Base |
| Sodium hydroxide | NaOH(aq) | Drain deaner | POISON | Blue | Yes | Base |

- 1. Refer to Model 1.
 - a. What is the common chemical name for Vitamin C?

Ascorbic Acid

& Is Vitamin C classified as an acid or a base?

Acid

Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.

Sour, turns litmus paper red, conducts electricity

3. Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?

H at the beginning

 Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

Turns litmus paper blue and conducts electricity

Acids And Bases Pogil Answer Key

E Staff

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