# What Is a Buffer in Chemistry?

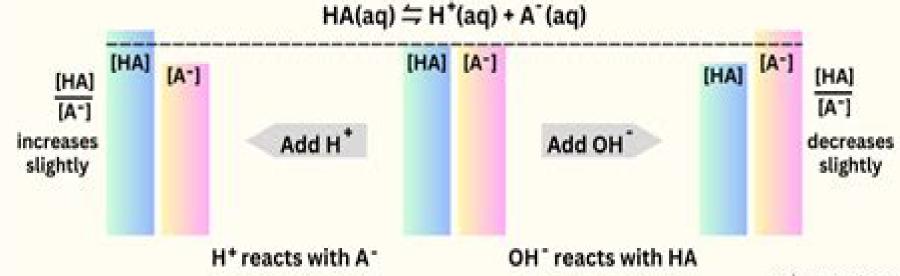
A buffer is a solution that resists changes in pH when a small amount of acid or base is added.

A buffer solution consists of a weak acid and its conjugate base or a weak base and its conjugate acid.

$$HA(aq) + OH^{-}(aq) \rightarrow A^{-}(aq) + H_{2}O(l)$$
  
 $HA(aq) + H_{2}O(l) \leftarrow A^{-}(aq) + H^{+}(aq)$ 

$$[H^+] = \frac{[HA]}{[A^-]}$$

$$HA(aq) = H^*(aq) + A^*(aq)$$



# **Buffers In Household Products**

# **Roman Wölfel**

**Buffers In Household Products:** 

The Enigmatic Realm of **Buffers In Household Products**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Buffers In Household Products** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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