#### Guided Notes: Matter and Chemistry Introduction

- Chemistry is the study of matter and its properties as well as how matter changes.
- Matter is anything that has mass and takes up space.

#### Properties of Matter

 A physical property is a characteristic of a substance that can be observed without changing the substance's composition. Examples include:

le.g., color, shape, texture)
(e.g., melting point, boiling point)

(e.g., density, solubility)

- A chemical property is a characteristic of a substance that describes its ability to change into other substances.

  Examples include:
  - (e.g., burning, rusting)

#### States of Matter

- has a definite shape and volume.

  Particles are tightly packed and vibrate in place.
- has a definite volume but not a definite shape. Particles are close together but can move around.
- has no definite shape or volume.

  Particles are far apart and move around randomly.
- exists at extremely high temperatures; particles are ionized.

#### Changes of Matter

 A physical change alters the form or appearance of a substance but does not change its composition. Examples include:

(e.g., cutting, tearing)

e \_\_\_\_\_\_ (e.g., melting, freezing, boiling)
e \_\_\_\_\_\_ (e.g., dissolving)

 A chemical change results in the formation of new substances with different properties. Examples include:

(e.g., burning, cooking)
 (e.g., rusting, tarnishing)

# Elements and Compounds

# **Answers To The Chemistry Guided Work**

Jo A. Beran

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