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# ACS Organic Chemistry



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### **Acs Organic Chemitry Study Guide**

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web answer the following questions consider the following illustration draw the lewis dot structure for hcl if the penguin represents a hydrogen atom and the polar bear represents a chlorine atom what does the ice cream represent in the drawing what do you think the picture is trying to illustrate would hcl be attracted to the charged wand

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web 1 how does the comic strip define a polar molecule 2 define electronegativity as you understand it after reading the first two pages of the comic strip 3 what is the artist trying to represent by two polar bears arm wrestling or two penguins arm wrestling 4 what three types of bonds are represented on the third page of the comic strip

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web so the final representation of the nh 3 molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the nh 3 molecule

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web question using polar bears and penguins create an illustration showing an ammonia molecule nh3 hint you may wish to start with a lewis dot structure

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web polar bears and penguins purpose in this lesson you will be exploring polarity and bonding between atoms in greater

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detail a comic book will provide new information about these topics and will introduce you to the concept of electronegativity which helps us understand partial charges