

Acids and bases review sheet Answer key

1. List the differences between an acid and a base. Your description should include such things as whether both are electrolytes, which contains a hydrogen ion and which includes a hydroxide ion, and you should also include two examples of an acid and two examples of a base that are present in your household.

Acid	base
Conducts electricity Contains hydrogen ion Proton donor Turns litmus paper red React with most metals Releases hydronium Ex: lemon Vinegar Battery acid (sulfuric acid)	conducts electricity contains hydroxide ion proton acceptor turns litmus paper blue releases hydroxide ion Ex: Windex bleach soap

2. Listed below are some of the properties of acids and bases. Fill in the blanks with the appropriate word, **acids** or **bases**:

- a) **A** produce hydrogen ions (H^+) in solution
- b) **B** have a bitter taste
- c) **B** produce hydroxide ions (OH^-) ions
- d) **A** react with bases to form salts
- e) **A** have a sour taste
- f) **B** have a slippery, soapy feel
- g) **B** react with acids to form salts
- h) **A** react with many metals

3. Write an equation for the dissociation of the following **acids** in water: follow example

- a) $\text{HClO}_4 \rightarrow \text{H}^+ + \text{ClO}_4^-$
- b) $\text{H}_2\text{SO}_4 \rightarrow 2\text{H}^+ + \text{SO}_4^{2-}$
- c) $\text{HC}_2\text{H}_3\text{O}_2 \rightarrow \text{H}^+ + \text{C}_2\text{H}_3\text{O}_2^-$
- d) $\text{H}_2\text{S} \rightarrow 2\text{H}^+ + \text{S}^{2-}$
- e) $\text{HCl} \rightarrow \text{H}^+ + \text{Cl}^-$
- f) $\text{HNO}_3 \rightarrow \text{H}^+ + \text{NO}_3^-$

4. Write an equation for the dissociation (ionization) of the following **bases** in water: (follow example)

- a) $\text{Mg}(\text{OH})_2 \rightarrow \text{Mg}^{2+} + 2\text{OH}^-$
- b) $\text{NaOH} \rightarrow \text{Na}^+ + \text{OH}^-$
- c) $\text{Ba}(\text{OH})_2 \rightarrow \text{Ba}^{2+} + 2(\text{OH}^-)$
- d) $\text{KOH} \rightarrow \text{K}^+ + \text{OH}^-$
- e) $\text{Mg}(\text{OH})_2 \rightarrow \text{Mg}^{2+} + 2(\text{OH}^-)$
- f) $\text{LiOH} \rightarrow \text{Li}^+ + \text{OH}^-$

5) A. What type of reaction is a **neutralization** reaction? **Double replacement**

B. Write the net equation for neutralization:



Acid And Base Test Review Sheet Answers

Margaret G. Kimble



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web bir rachel carson eseri olan silent spring en cazip fiyat ile d r de keşfetmek için hemen tıklayınız

silent spring wikipedia - Sep 18 2023

web silent spring is an environmental science book by rachel carson published on september 27 1962 the book documented the environmental harm caused by the indiscriminate use of pesticides carson accused the chemical industry of spreading disinformation and public officials of accepting the industry's marketing claims unquestioningly