

SECTION 14.2

Acid-Base Theories

Teacher Notes and Answers

SECTION 2 ACID-BASE THEORIES

1. Brønsted-Lowry acids can donate protons to solutions other than water solutions.
2. Brønsted-Lowry bases accept donated protons. Arrhenius bases add hydroxide ions to aqueous solutions.
3. H_3O^+ , H_2PO_4^- , HPO_4^{2-} , PO_4^{3-}
4. The sodium ion is not a Lewis acid because it does not form a covalent bond. It accepts a single electron to form an ionic bond.
5. base; acid
6. Lewis acids

Review

1. Monoprotic acids can only donate one proton per molecule. Diprotic acids can donate two protons per molecule. Triprotic acids can donate three protons per molecule.
2. H_2CO_3 : proton donor, acidic; H_2O : proton acceptor, basic
3. AlCl_3 : electron pair acceptor, Lewis acid; H_2O : electron pair donor, Lewis base

- 4a. NaOH is an Arrhenius base because it increases the hydroxide ion concentration when it dissociates in water.
- 4b. H_2O is a Brønsted-Lowry base because it is a proton acceptor. It is also a Lewis base because it donates an electron pair to a hydrogen ion to form a hydronium ion.
- 4c. NH_3 is a Brønsted-Lowry base because it is a proton acceptor. It is also a Lewis base because it donates an electron pair to a hydrogen ion to form an ammonium ion.

Acid Base Theories Section Review Answers

**Thirumalaisamy P.
Velavancorresponding**



Acid Base Theories Section Review Answers:

Reviewing **Acid Base Theories Section Review Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Acid Base Theories Section Review Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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