

CHAPTER 26 : ACID-BASE

Total body water volume: _____ this is 60% of body weight

Water: universal solvent

Solutes: nonelectrolytes and electrolytes

Extracellular and intracellular

- ECF : the fluid on the _____ the cell major cation is sodium and major anion is chloride
- Sodium is the most abundant cation in the extracellular fluid
- ICF: fluid _____ the cell major cation is potassium

Water intake is _____ to water output 2500 ml/day

Water output: urine, insensible water loss, perspiration, and feces

Water reabsorption: in collecting ducts proportional to ADH release

Aldosterone: important in regulation of sodium ion concentration

*water follows sodium

Decrease ADH: diluted urine and decrease in volume in body fluids

Increase ADH: concentrated urine

Atrial natriuretic peptide: decrease BP and blood volume

Important buffer systems

- Respiratory system
- Kidneys

normal pH of body fluids

- Arterial blood = pH 7.4
- Venous blood and IF fluid = pH 7.35
- ICF = pH 7.0
- Alkalosis or alkalemia = arterial blood pH > 7.45
- Acidosis or acidemia = arterial pH < 7.35
- Physiological acidosis = pH between 7.0 - 7.35

Hydrogen ion regulation

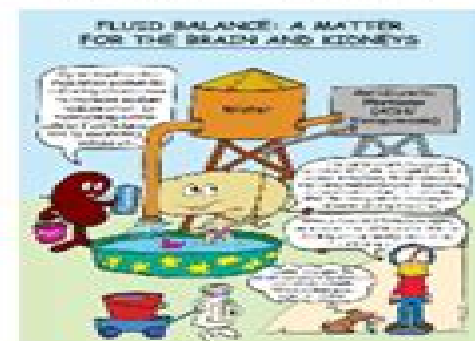
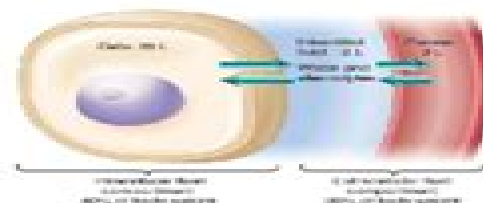
1. Chemical buffer system _____ (rapid) first line of defense
2. Respiratory centers - in brain stem act within 1 - 3 min
3. Renal mechanisms - _____ most potent

Three major chemical buffers

Chemical buffer: resist change in pH changes when a strong acid or base is added

1. Bicarbonate buffer system
2. Phosphate buffer system
3. Protein buffer system

Major fluid compartments in the body



Acid Base Study Guide Answers

James E. Brady, Fred Senese



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