

ACIDS AND BASES WS 5

Question 13) Write the formulas of the acid and the base that formed the following salts.

CH_3COONa

CuSO_4

KClO_3

$\text{Al}_2(\text{SO}_4)_3$

NH_4Cl

$\text{Ba}(\text{NO}_3)_2$

NH_4NO_3

CH_3COONa : CH_3COOH and NaOH

CuSO_4 : H_2SO_4 and $\text{Cu}(\text{OH})_2$

KClO_3 : HClO_3 and KOH

$\text{Al}_2(\text{SO}_4)_3$: $\text{Al}(\text{OH})_3$ and H_2SO_4

NH_4Cl : NH_4OH and HCl

$\text{Ba}(\text{NO}_3)_2$: $\text{Ba}(\text{OH})_2$ and HNO_3

NH_4NO_3 : NH_3 and HNO_3

Question 14) Write two observations you would make when quicklime is added to water.

Question 15) What are the uses of Bleaching powder?

- (i) It is used for bleaching cotton and linen in the textile industry, for bleaching wood pulp in paper factories and for bleaching washed clothes in laundry
- (ii) It is used as an oxidising agent in many chemical industries
- (iii) It is used for disinfecting drinking water to make it free of germs

Question 16) State the chemical property in each case on which the following uses of baking soda are based :-

as an antacid

as a constituent of baking powder.

Question 17) How is plaster of Paris obtained? What reaction is involved in the setting of a paste of plaster of Paris?

Question 18) Acids show their properties only in the presence of water? explain?

Question 19) A weak acid is added to a concentrated solution of hydrochloric acid. Does the solution become more or less acidic?

Question 20) Write the chemical name and formula of washing soda. What happens when

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SA Dillow



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