# WORKSHEET

# Acid-base Titration I Maximum score: 100 points

Your Name: Partner: Lab Section (Day & Time):

 Attach your annotated plots and label the equivalence point and the half-equivalence point appropriately for the titration curves for HCL KHP, and one trace for the unknown acid, and record the data from the plots.
 (30 points)

Acid	presentation	pK, or pH at hair, equivalence point	pH at equivalence point	pH at end of plot
HCI	2 + 2	XXXXXXXXXX	5.3	11.75
HORSE	4.6	5	16.5	11.70
Spinston Comments	2.3	2.9	9.0	11.600

2. Complete the table using data from your lab notebook. Calculate the acid concentration using  $M_{\pi}F_{\pi}=M_{\pi}F_{\pi}$ . The unknotes acid will require a correction for the distribution. (20 pm)

Acid Triol	Wolume send one;	Concentration (M)	Working of bases at Equivalence point (ret. NaOH)		Calculated Acad Concentration
MOHP District Orionows Tried 1	55	- 1117	막 시크 건 시트니	.1188	-0089
and Granown	er ele	-1111		XXXXXXXXX	.0051

# **Acid Base Titration Worksheet Chemistry If8766 Answers**

**Marco Cascella** 

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