







Enhanced dipole interaction between bonded H

"structure" to liquid water, solubilizes alcohols,

 Hydrophobic = Lipophillic: Repelled by polar group; insoluble in water, affinity for nonpolar

2. Hydrophilic = Lipophobic: Affinity for polar

group; soluble in water, repelled by nonpolar

fatty acids, artifices, sugars, and amino acids

B. Types of Chemical Groups

EX: alcohol, amine, carboxylic acid

EX: alleane, urene, alleene

and the lone-pair of neighboring O. N or S; gives

layers: compounds are partitioned between the layers based on chemical properties (acid/base, polar, nonpolar, ionici)

3. Physical principles
a Colligative properties depend on solvent identity and concentration of solute; a solution has a higher bedling point, lower freezing point and lower vapor pressure than the pure solvent

peids, amino acids and nucleotides.

C. Behavior of Solutions

and lower vapor pressure than the pure solvent b. Biochemical example — Osmotic Pressure: Water diffuses through a semi-permeable membrane from a hypotonic to a hypotonic region; the flow predaces a force, the comotic pressure, on the hypothesic side

3. Amphipaties Polar and nonpolar functionality.

common for most biochemical molecules: fatty

Miscibles 2 or more substances form 1 phase;

occurs for polar + polar or nonpolar + nonpolar

2. Immiscible: 2 liquids form aqueous and organic

## Osmotic Pressure

He Osmotic pressure (in atm)

is Van't Hoff factor - it of sons per solute molecule.

Ms Solution molarity (moley/L)

Re Gas constant = 0.082 L atm mol 1 K 1

Tr Absolute temperature (in Kelvin)

#### 4. Solutions of gases.

- Henry's Law: The amount of gas dissolved in a liquid in proportional to the partial pressure of the gas
- Carbon dioxide dissolves in water to form carbonic acid
- Oxygen is carried by hemoglobin in the blood d Pollutants and toxins dissolve in bodily fluids;
- react with tissue and interfere with reactions EX; Sulfur oxides and nitrogen oxides yield acids; oxides exidince lung tissue; hydrogen cyanide disables the oxidation of electore

# **Biochemistry Quickstudy Academic**

**Denise Guinn** 

**Biochemistry Quickstudy Academic:** 

Unveiling the Magic of Words: A Report on "Biochemistry Quickstudy Academic"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Biochemistry Quickstudy Academic**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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