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by its rate law for example the rate of the gas phase decomposition of dinitrogen pentoxide $2\text{N}_2\text{O}_5 \rightarrow 4\text{NO}_2 + \text{O}_2$ has been found to be directly proportional to the concentration of N_2O_5 text rate k n

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see table below notice that the starting concentrations of ce no and ce h

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