

Name _____ answer key _____ Date _____

Stoichiometry: Mass – Mole Worksheet

Based on the following chemical equation, please answer the following questions!



Calculate the amount of moles of NaCl needed to produce 75 g of FeCl₃.

9 FeCl ₃	1 mole FeCl ₃	mole NaCl
	9 FeCl ₃	mole FeCl ₃

Calculate the amount of Fe₂(SO₄)₃ in grams produce 215 g Na₂SO₄.

9 Na ₂ SO ₄	1 mole Na ₂ SO ₄	mole Fe ₂ (SO ₄) ₃	9 Fe ₂ (SO ₄) ₃
	9 Na ₂ SO ₄	mole Na ₂ SO ₄	1 mole Fe ₂ (SO ₄) ₃

Calculate the amount in grams of FeCl₃ produced with 95g of Fe (SO₄)₃

9 Fe ₂ (SO ₄) ₃	1 mole Fe ₂ (SO ₄) ₃	mole FeCl ₃	9 FeCl ₃
	9 Fe ₂ (SO ₄) ₃	mole Fe(SO ₄) ₃	1 mole FeCl ₃

Calculate the amount of moles of Fe₂(SO₄)₃ needed to react with 123.5 g of NaCl

9 NaCl	1 mole NaCl	mole Fe(SO ₄) ₃
	9 NaCl	mole NaCl

Based on the following chemical equation, please answer the following questions!



9 Ca(HCO ₃) ₂	1 mole Ca(HCO ₃) ₂	mole CaCO ₃
	9 Ca(HCO ₃) ₂	mole Ca(HCO ₃) ₂

Answers For Study Guide Mass And Moles

Leo J. Malone, Theodore O. Dolter



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Table of Contents Answers For Study Guide Mass And Moles

1. Understanding the eBook Answers For Study Guide Mass And Moles
 - The Rise of Digital Reading Answers For Study Guide Mass And Moles
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Study Guide Mass And Moles
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Study Guide Mass And Moles
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Study Guide Mass And Moles
 - Personalized Recommendations
 - Answers For Study Guide Mass And Moles User Reviews and Ratings
 - Answers For Study Guide Mass And Moles and Bestseller Lists
5. Accessing Answers For Study Guide Mass And Moles Free and Paid eBooks
 - Answers For Study Guide Mass And Moles Public Domain eBooks
 - Answers For Study Guide Mass And Moles eBook Subscription Services
 - Answers For Study Guide Mass And Moles Budget-Friendly Options

6. Navigating Answers For Study Guide Mass And Moles eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Study Guide Mass And Moles Compatibility with Devices
 - Answers For Study Guide Mass And Moles Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Study Guide Mass And Moles
 - Highlighting and Note-Taking Answers For Study Guide Mass And Moles
 - Interactive Elements Answers For Study Guide Mass And Moles
8. Staying Engaged with Answers For Study Guide Mass And Moles
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Study Guide Mass And Moles
9. Balancing eBooks and Physical Books Answers For Study Guide Mass And Moles
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Study Guide Mass And Moles
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For Study Guide Mass And Moles
 - Setting Reading Goals Answers For Study Guide Mass And Moles
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Study Guide Mass And Moles
 - Fact-Checking eBook Content of Answers For Study Guide Mass And Moles
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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