

- [illegible]

CONCLUSIONS

¹⁰ Even the fact that the government is not required to disclose the properties of the underlying assets

[illegible]

think about this

think about this



...the same can be seen in the way that the...
...the same can be seen in the way that the...
...the same can be seen in the way that the...

ON YOUR OWN

2. Select, of the following additional processes, one that you think will be most important in the future. Explain the reasons.

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- 8.2. Which of the following substances particles are found in the nucleus, protons, electrons and neutrons? Which is the other one being?

THE PERIODIC TABLE OF THE ELEMENTS

THE PERIODIC TABLE OF THE ELEMENTS

The periodic table of the elements, Figure 2.1, is a tabular arrangement of the elements that is based on their atomic number (Z), which is the number of protons in the nucleus of an atom. The elements are arranged in rows and columns such that elements with similar chemical and physical properties are placed in the same group or column. The periodic table is a fundamental tool in chemistry, used to predict the properties of elements and to understand the trends in chemical behavior.

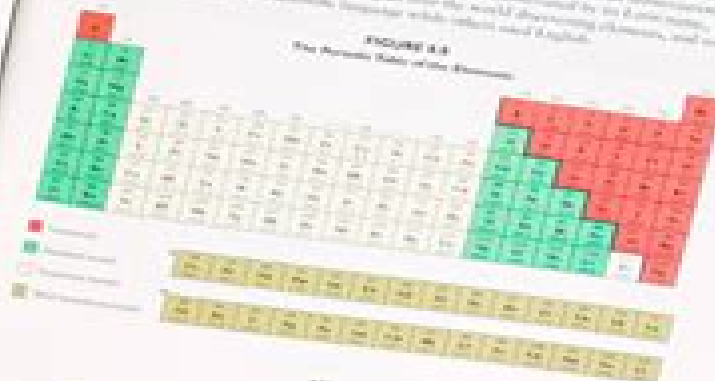
The periodic table is divided into four main sections: the s-block (groups 1 and 2), the p-block (groups 13-18), the d-block (transition metals, groups 3-10), and the f-block (lanthanides and actinides, groups 14-17). The elements are arranged in order of increasing atomic number, with hydrogen (Z=1) at the top left and the heaviest elements at the bottom right.

The periodic table is a powerful tool for understanding the periodic trends in chemical properties. For example, the electronegativity of elements increases from left to right across a period and decreases from top to bottom down a group. The atomic radius of elements increases from left to right across a period and decreases from top to bottom down a group.

The periodic table is also used to predict the chemical behavior of elements. For example, elements in the same group have similar chemical properties. Elements in the same period have similar physical properties.

The periodic table is a fundamental tool in chemistry, used to predict the properties of elements and to understand the trends in chemical behavior.

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