Diffusion

Particles move about randomly, and after a bit they end up evenly spaced. It's not rocket science, is it...

Don't Be Put Off by the Fancy Word

- "Diffusion" is simple. It's just the gradual movement of particles from places where there are lots of them to places where there are fewer of them — it's just the natural tendency for stuff to spread out.
- 2) Unfortunately you also have to learn the fancy way of saying the same thing, which is this:

DIFFUSION is the SPREADING OUT of particles from an area of HIGHER CONCENTRATION to an area of LOWER CONCENTRATION.

- Diffusion happens in both <u>solutions</u> and <u>gases</u> that's because the particles in these substances are free to <u>move about randomly</u>.
- 5) The simplest tupe is when different gases diffuse through each other. This is what's happening when the smell of perfume diffuses through the air in a room:



- 6) The <u>bigger</u> the <u>concentration gradient</u> (the <u>difference</u> in concentration), the <u>faster</u> the diffusion rate.
- A higher temperature will also give a faster diffusion rate because the particles have more energy, so move around faster.

Cell Membranes Are Kind of Clever ...

- They're clever because they hold the cell together <u>BUT</u> they let stuff in and out as well.
- 2) Dissolved substances can move in and out of cells by diffusion.
- Only very small molecules can diffuse through cell membranes though
 — things like coupen (needed for respiration see page 62), glucose, amino acids and water.
- 4) Big molecules like starch and proteins can't fit through the membrane:



- 5) Just like with diffusion in air, particles flow through the cell membrane from where there's a higher concentration (a lot of them) to where there's a lower concentration (not such a lot of them).
- 6) They're only moving about <u>randomly</u> of course, so they go <u>both</u> ways but if there are a lot <u>more</u> particles on one side of the membrane, there's a <u>net</u> (overall) movement <u>from</u> that side.
- 7) The larger the surface area of the membrane, the faster the diffusion rate, because more particles can pass through at once see page 23.

Revision by diffusion - you wish...

Wouldn't it be great if all the ideas in this book would just gradually drift across into your mind...

- Q1 A student adds a drop of ink to a glass of cold water.
 - a) What will the student observe to happen to the drop of ink. Explain your answer.
 - b) How might the observation differ if the ink was added to a glass of warm water?

[2 marks] [1 mark]

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Jessica J Manson

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