

- [illegible]

CONCLUSIONS

¹⁰ Even the most well-meaning entrepreneur is convinced that the world is full of unfairness.

[illegible]

think about this



There, at the time of the meeting of the board, someone there observed that the company was in a "very tight spot" and that the company was in a "very tight spot" and that the company was in a "very tight spot".

ON YOUR OWN

1. Select, of the following additional exercises are done:
2. Describe the effect of the other, and

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- 8.2 Explain the difference between an atom and an element.

THE PERIODIC TABLE OF THE ELEMENTS
The periodic table is a chart that lists all the elements known to science. It is organized into rows and columns based on their chemical and physical properties.

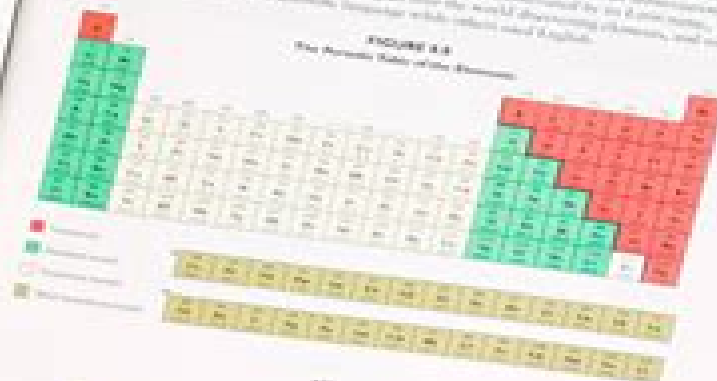
THE PERIODIC TABLE OF THE ELEMENTS

The periodic table of the elements (Figure 2.5) is a collection of atoms arranged based on their atomic number, which is the number of protons in the nucleus. Elements are arranged in rows and columns based on their chemical and physical properties. Elements in the same row (period) have similar chemical properties, while elements in the same column (group) have similar physical properties. The periodic table is divided into four main sections: s-block, p-block, d-block, and f-block. The s-block contains elements with one or two valence electrons, the p-block contains elements with three to six valence electrons, the d-block contains elements with seven to ten valence electrons, and the f-block contains elements with eight to ten valence electrons. The periodic table is a fundamental tool for understanding the properties and behavior of elements.

FIGURE 2.5

The periodic table of the elements, showing the arrangement of elements based on their atomic number and chemical properties. The table is divided into four main sections: s-block, p-block, d-block, and f-block. The s-block contains elements with one or two valence electrons, the p-block contains elements with three to six valence electrons, the d-block contains elements with seven to ten valence electrons, and the f-block contains elements with eight to ten valence electrons.

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