

2. (A) On the basis of CFT, discuss high spin octahedral complexes with examples. Why are tetrahedral complexes only high spin? 4

(B) What is ion exchange chromatography? Discuss in detail the experimental technique used in the separation of ions. 4

OR

2. (C) Calculate spin magnetic moment of $[\text{Fe}(\text{CN})_6]^{4-}$ and $[\text{Fe}]^{2+}$ using CFT. 2

(D) Discuss how orbital motion contributes towards magnetic moment in octahedral complexes. 2

(E) Explain the principle of:

(i) Adsorption chromatography and

(ii) Partition chromatography. 2

(F) Calculate R_f value for Co^{2+} and Ni^{2+} ions if distance travelled by solvent, Co^{2+} and Ni^{2+} are 9.8 cm, 7.8 cm and 0.6 cm respectively. 2

3. (A) Draw flow sheet diagram of double beam spectrophotometer. Discuss application of colorimeter for the estimation of $\text{Cu}(\text{II})$ as copper ammonia complex. 4

(B) How does the stability of complex depend on:

(i) Charge and size of the metal ion and

(ii) Basicity of the ligands. 4

3. (C) State Beer-Lambert's Law. Give its deviation. 2

(D) A cell of thickness 0.1 cm contains $5 \times 10^{-2} \text{ M}$ solution of a solute. If the solution has absorbance 0.85, calculate molar extinction coefficient. 2

(E) Describe Job's method for the determination of composition of $\text{Fe}(\text{III})$ SSA complex. 2

(F) Explain why stability constant value decreases in the order:



4. (A) What are organometallic compounds? Discuss ionic and covalent organometallic compounds with suitable examples. 4

(B) (i) What is the action of heat on:

(a) $\text{Ni}(\text{Co})_4$ and (b) $\text{Fe}(\text{Co})_5$?

(ii) Write IUPAC names of the following compounds:

(a) $\text{C}_6\text{H}_5\text{MgCl}$ and (b) $(\text{C}_6\text{H}_5)_2\text{Zn}$. 4

OR

4. (C) Discuss the structure and bonding in Zeise's salt. 2

(D) How is organoaluminium compound prepared from grignard reagent? Draw the structure of $\text{Al}_2(\text{CH}_3)_6$. 2

(E) Write two uses of organotin compound. 2

(F) What is meant by back- π -bonding? Explain this concept in metal carbonyls. 2

5. (A) Discuss:

(i) Metalloporphyrins and

(ii) Calcium pump. 4

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