

ADP TIME CLOCK INSTRUCTIONS

Job Classification: Hourly Employee (multiple jobs or sites)

As you are eligible and likely to perform 1 or more jobs in 1 or more locations, you will need to specify the location and job code of your work at the beginning of your work day. The sign-in instructions are illustrated below. You will also need to perform this procedure if you are changing your location or job during the day. To sign-out at the end of your work day, simply swipe your card at the time clock.



Step 1: Press the Transfer button on the left side of the time clock



Step 2: Using the numeric keypad, enter the 2-digit location code of your building



Step 3: Press the Enter button (the time clock will prompt for your 3-digit job code)



Step 4: Enter the 3-digit job code for the work you are performing.



Step 5: Press the Enter button (the time clock will prompt for your employee ID)



Step 6: Swipe your card (A successful read will generate a beep and a green light)

Adp 4500 Time Clock User Manual

DJ Losen



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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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