

Acids, Bases and Buffers MCQ Test With Questions and Answers

Which of the following statements is CORRECT?

- A. An acid is an electron donor and a base is an electron acceptor.
- B. An acidic solution has a pH less than 7 and a basic solution has a pH greater than 7.
- C. Neutralization of a strong acid by a strong base gives only water.
- D. The pH of saliva is normally in the range of 8.5-9.5 - answerB

Which statement about buffers below is most correct?

- A. A buffer is any acid and base which together control the concentration of pH in the blood.
- B. A buffer is the solution which allows CO₂ to be lost from the lungs in order to control the pH of the lungs.
- C. A buffer is a weak base and its acid salt or a weak acid and its basic salt. The ratio of the two components helps maintain blood pH levels.
- D. A buffer is a mixture of two acids which together help to maintain blood pH - answerC

Which of the following statements is FALSE?

- A. An acid is a proton donor and a base is a proton acceptor.
- B. An acidic solution has a pH greater than 7 and a basic solution has a pH less than 7.
- C. Neutralization of an acid by a base gives a solution of salt in water.
- D. The pH of the stomach is normally in the range of 1.6-1.8 - answerB

Neutralization of an acid by a base can be represented by the following equation:



This means that the pH of the solution after the neutralization is approximately:

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