

Guided Practice - Acids & Bases**Properties of Acids and Bases**

1. Match the following terms with their descriptions:

- | | | |
|------------------------|-------|--|
| a. Arrhenius Acid | _____ | the substance formed when a proton is added to _____ |
| b. Arrhenius Base | _____ | _____ a base |
| c. Conjugate Acid | _____ | the remaining substance when a proton is lost from an acid |
| d. Conjugate Base | _____ | a substance that produces OH^- in solution |
| e. Bronsted-Lowry Acid | _____ | a substance producing H^+ in solution |
| f. Bronsted-Lowry Base | _____ | a proton acceptor |
| | | a proton donor |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are H_2O conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- HSO_4^- , SO_4^{2-}
- HBr , BrO^-
- H_3PO_4 , PO_4^{3-}
- H_2O , HO^-

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

Acids And Bases Guided Study Workbook

Neil E. Schore



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Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14

Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Chemistry for the IB Diploma Study and Revision Guide Christopher Talbot, Richard Harwood, 2017-07-24 Exam Board IB Level IB Subject Chemistry First Teaching September 2014 First Exam Summer 2016 Stretch your students to achieve their best grade with these year round course companions providing clear and concise explanations of all syllabus requirements and topics and practice questions to support and strengthen learning Consolidate revision and support learning with a range of exam practice questions and concise and accessible revision notes Practise exam technique with tips and trusted guidance from examiners on how to tackle questions Focus revision with key terms and definitions listed for each topic sub topic

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Organic Chemistry, 12e Binder

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