

Guided Practice - Acids & Bases

Properties of Acids and Bases

1. Match the following terms with their descriptions:

- | | | |
|------------------------|-------|--|
| a. Arrhenius Acid | _____ | the substance formed when a proton is added to _____ |
| b. Arrhenius Base | _____ | _____ a base |
| c. Conjugate Acid | _____ | the remaining substance when a proton is lost from an acid |
| d. Conjugate Base | _____ | a substance that produces OH^- in solution |
| e. Bronsted-Lowry Acid | _____ | a substance producing H^+ in solution |
| f. Bronsted-Lowry Base | _____ | a proton acceptor |
| | | a proton donor |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are H_2O conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- HSO_4^- , SO_4^{2-}
- HBr , BrO^-
- H_3PO_4 , PO_4^{3-}
- H_2O , HO^-

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

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web oct 31 2023 updated 8 36 pm pst october 30 2023 taipei taiwan ap three chinese astronauts returned to earth on tuesday morning after six months aboard china s orbiting space station jing haipeng zhu yangzhu and gui haichao emerged in good health from the return capsule near the jiuquan satellite launch center on the edge of

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web establishment of a complete end to end ecosystem comprising the starlab space station as a low earth orbit destination and a potential esa developed european transportation system cargo and crew leveraging standardised interfaces in the interest of an open access policy this agreement reflects esa s ambition to enable a smooth

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