

Exam 2 - Cell Biology Cheat Sheet

by E_Hilliker via cheatography.com/182438/cs/37955/

Common Fr	unctional G	токра
Htydroxyl	-OH	poter
Carboxyl	COOH	polar
Carbonyl	-00	polar
Armino.	-04113	polar
Phosphate	-P03	polar
Methyl	-CH3	mon- polar
Suffhydryl	-SIH	mon-

Organic Compounds Overview

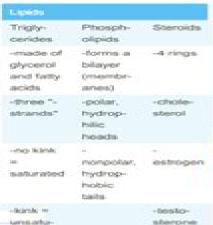
good-ar-



"more details underneath this belowik.

Controllyd	Continue Co	
Monosa och- arides	Disaco- harides	Polysa- con- arides
glucose	-sucrose	"starch"
fructose	-lactose	cellulose
	-dextrose	-ichition .
	-mattorse	arrytose
	galactose	gfycogen

"sugars can dissolve in H20. because they are NON-POLAR!



Non-ionized form of amino acid



"R groups can be a mix of the following: uncharged, non-polar, polar, hydrophilic, hydrophobic

"amino acids are joined by a PEPTIDE BOND

"this results in a dehydration. (talking H210 out)

"protein - shape

"amount of Pi (phosphate group) conn athemses

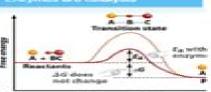
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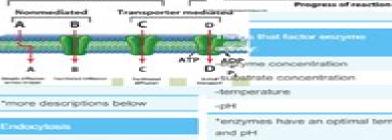
Osmosis		
Hypertonic	Hypotonic	Isiotonio
high	low.	equili-
solute. less water	solute. high	brium
	water	
cell shriveled	cell swells (hippo)	cett is normal

Paradise

-bind to a substrate. These fit depending on the SHAPE of enzymes.

-competitive inhibitors can block substrates by changing the shape of the enzyme.





*enzymes have an optimal temp.

"this depends on the specific enzymet

"enzymes usually can go back to its original form, unless it DENATURES



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Michael Seilmaier

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