# Acid-Base Bafance

O-6.9

Acidic

Alkaline

pH of 7 = Neutral (Water)

pH of Blood	Kanges 7,35-7.45
ETCO2	35-45mmHg
HCO3 (Bicarbonate)	22-26 meq/L

HCO3: Normally, O2 combines with glucose to create energy and CO2 is an acid waste by product of this. Bicarbonate is a buffer produced by the body that is alkaline, and binds with CO2 to neutralize it to allow the body to get rid of it.

# **Acid And Base Reactions Study Guide Answers**

John Krenos, Joseph Potenza, Loretta
Jones, Lynn Koplitz, Thomas Spence

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