

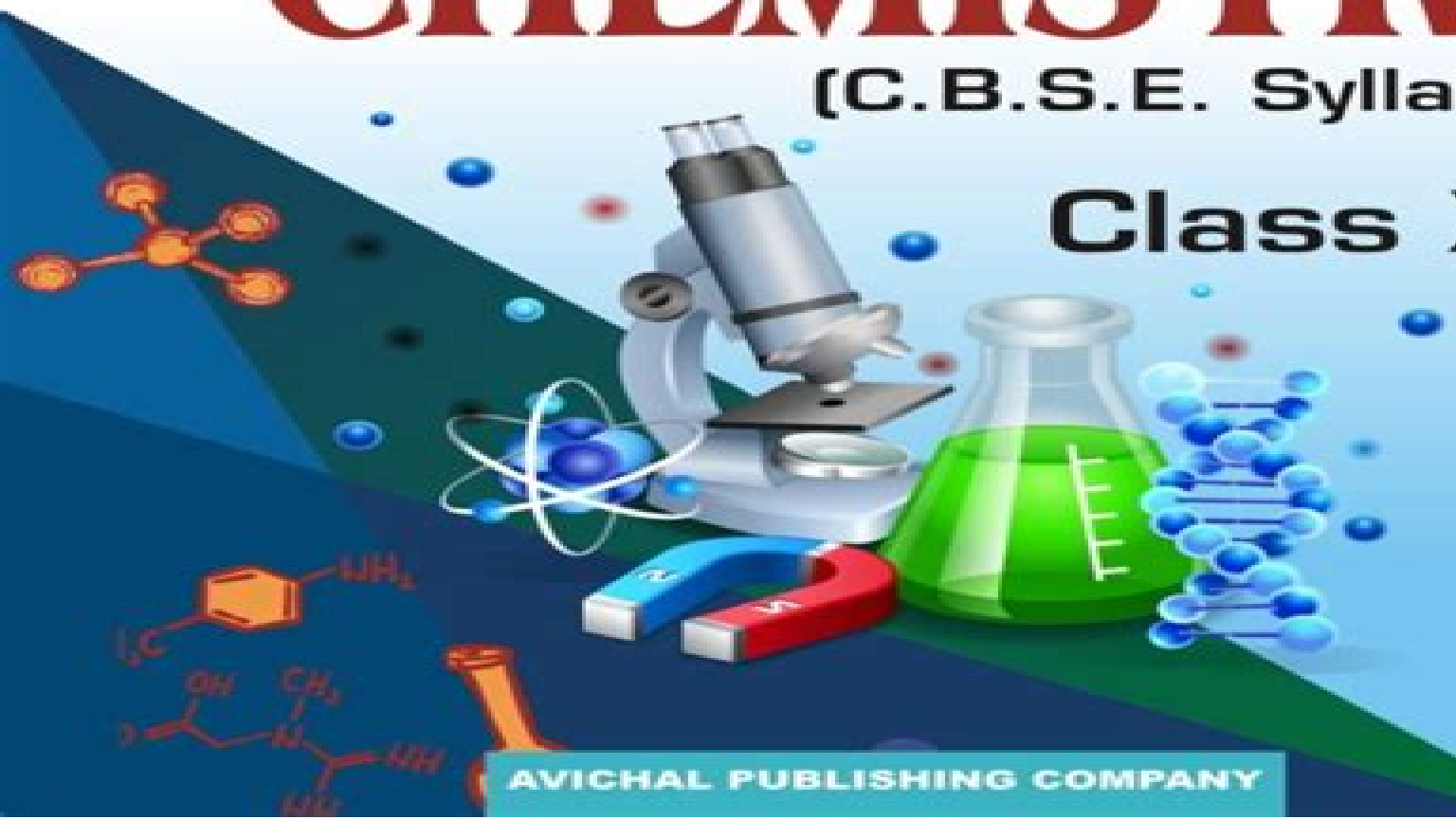
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# LABORATORY MANUAL CHEMISTRY

(C.B.S.E. Syllabus)

Class XI



AVICHAL PUBLISHING COMPANY

# Apc Chemistry Lab Manual Class 11

**Neena Sinha, R Rangarajan, R P  
Manchanda, R K Gupta, Rajesh Kumar**



## **Apc Chemistry Lab Manual Class 11:**

Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

**1 Basic Laboratory Techniques**

1 To cut a glass tube or glass rod  
2 To bend the glass rod at an angle  
3 To draw a glass jet from a glass tube  
4 To bore a cork and fit a glass tube into it  
**Viva Voce**

**2 Characterisation and Purification of Chemical Substances**

1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique  
**Viva Voce**  
2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method  
**Viva Voce**  
3 To prepare crystals of pure potash alum  $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$  from the given impure sample  
4 To prepare the pure crystals of copper sulphate from the given crude sample  
5 To prepare pure crystals of benzoic acid from a given impure sample  
**Viva Voce**  
3 **Measurement of pH Values**

1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper  
2 To determine and compare the pH values of solutions of strong acid HCl and weak acid  $CH_3COOH$  of same concentration  
3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper  
4 To study the pH change by common ion  $CH_3COO^-$  ion in case of weak acid  $CH_3COOH$   
5 To determine the change in pH value of weak base  $NH_4OH$  in presence of a common ion  $NH_4^+$   
**Viva Voce**

**4 Chemical Equilibrium**

1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions  
2 To study the shift in equilibrium between  $Co(H_2O)_6^{2+}$  and  $Cl^-$  ions by changing the concentrations of either of the ions  
**Viva Voce**

**5 Quantitative Analysis**

1 To prepare M 10 oxalic acid solution by direct weighing method  
2 To prepare M 10 solution of sodium carbonate by direct weighing method  
3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid  
4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution  
**Viva Voce**

**6 Qualitative Analysis**

**Analysis of Anions**  
**Analysis of Cations**  
**Viva Voce**

**7 Detection of Elements in Organic Compounds**

1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test  
2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test  
**Viva Voce**

**INVESTIGATORY PROJECTS**

**1 Checking of Bacterial Contamination in Water**

1 To check the bacterial contamination in drinking water by testing sulphide ions  
**Viva Voce**

**2 Methods of Water Purification**

1 To purify water from suspended impurities by using sedimentation  
2 To purify water by boiling  
3 To purify water by distillation method  
4 To purify water by reverse osmosis technique  
5 To purify water by GAC method  
6 To purify water by bleach treatment  
7 To purify water by oxidising agent  
8 To purify water by ozone treatment method  
**Viva Voce**

**3 Water Analysis**

1 To test the hardness of different water samples  
**Viva Voce**

**4 Foaming Capacity of Various Soaps**

1 To compare the foaming capacity of different washing soaps  
2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap  
**Viva Voce**

**5 Tea Analysis**

1 To study the acidity of different samples of tea leaves tea by using pH paper  
**Viva Voce**

**6 Analysis of Fruits and Vegetable Juices**

1 To analysis the fruit and vegetable juices for the constituent present in them

Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce *Chemistry Lab Manual* Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual *Acp Chem 1172 General Chemistry Lab Manual* Brooks/Cole,2015-11-11 **Current Index to Journals in Education** ,1977 **Acp Chem 1171 General Chemistry Lab Manual** Brooks/Cole,2014-04-25 The Software Encyclopedia ,1986 **Popular Photography** ,1990-11 **Nutrition Abstracts and Reviews** ,1998 General Chemistry I Laboratory Manual, 11E - CHM 111 Brightpoint Community College,2023-05-26 **Literature on Information Retrieval and Machine Translation** Service Bureau Corporation,1959

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