

Chapter 2 Active Reading Guide

The Chemical Context of Life

This chapter covers the basics that you may have learned in your chemistry class. Whether your teacher goes over this chapter, or assigns it for you to review on your own, the questions that follow should help you focus on the most important points.

Section 1

1. Define and give an example of the following terms:

Term	Definition
matter:	All physical substances.
element:	Chemical substances that cannot be broken down
compound:	A mix of two or more elements.

2. What four elements make up 96% of all living matter?

Carbon, Oxygen, Hydrogen, and Nitrogen

3. What is the difference between an *essential element* and a *trace element*?

While essential elements are needed in large quantities for an organism to live a healthy life, trace elements are needed in very little amounts.

Section 2

4. Sketch a model of an atom of helium, showing the electrons, protons, and neutrons, and atomic nucleus.



5. What is the atomic number of helium? 2 Its atomic mass? 4

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M Mark



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Journal of the American Medical Association ,2008-07 **Best Practices in Gifted Education** Ann Robinson, Bruce M. Shore, Donna Enersen, 2021-09-03 A gifted education Legacy Award winner Best Practices in Gifted Education provides concise up to date research based advice to educators administrators and parents of gifted and talented youth The 29 practices included in this volume are the result of an extensive examination of educational research on what works with talented youth The interest in culturally diverse and low income learners the means to identify talents and the need for curriculum that appropriately challenges high ability youth constitute just a few of the 29 practices Each practice is organized into a chapter containing two sections What We Know and What We Can Do The first section briefly describes the practice and summarizes the research The second section suggests what course of action a parent teacher or administrator might take at home in the classroom or at school The book is a must have for those who want a guide that makes a connection between research and practical action in gifted education A service publication of the National Association for Gifted Children Washington DC This designation indicates that this book has been jointly developed with NAGC and that this book passes the highest standards of scholarship research and practice

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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution
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web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves
exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol rxn}$
write equations to similarly represent the dissolving process for

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