

A. Balance each of the following chemical equations.



B. Solve the following problems and write your solutions and answers in the space provided for.

Use the atomic weights in the periodic table of your procedure manual.

- i. Iron (II) sulfate, FeSO₄, is prescribed for the treatment of anemia.

- a) How many moles of FeSO₄ are present in a

Activity 3 8 Chemical Equations 1 Introduction Answers

JA Banks



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Kinetic Modeling of Reactions In Foods Martinus A.J.S. van Boekel, 2008-12-18 The level of quality that food maintains as it travels down the production to consumption path is largely determined by the chemical biochemical physical and microbiological changes that take place during its processing and storage Authored by an internationally respected food quality expert *Kinetic Modeling of Reactions in Foods* demonstrates how to effectively capture these changes in an integrative fashion using mathematical models Thus kinetic modeling of food changes creates the possibility to control and predict food quality from a technological point of view Illustrating how kinetic modeling can predict and control food quality from farm to fork this authoritative resource Applies kinetic models using general chemical physical and biochemical principles Introduces Bayesian statistics in kinetic modeling virtually uncharted territory in the food science field Integrates food science kinetics and statistics to predict and control food quality attributes using computer models Uses real world examples rather than hypothetical data to illustrate concepts This essential reference is an indispensable guide to understanding all aspects of kinetic food modeling Unlike many other kinetic volumes available this book opens the door to the many untapped research opportunities in the food science realm where mathematical modeling can be applied

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