

CHAPTER 26 : ACID-BASE

Total body water volume: _____ this is 60% of body weight

Water: universal solvent

Solutes: nonelectrolytes and electrolytes

Extracellular and intracellular

- ECF : the fluid on the _____ the cell major cation is sodium and major anion is chloride
- Sodium is the most abundant cation in the extracellular fluid
- ICF: fluid _____ the cell major cation is potassium

Water intake is _____ to water output 2500 ml/day

Water output: urine, insensible water loss, perspiration, and feces

Water reabsorption: in collecting ducts proportional to ADH release

Aldosterone: important in regulation of sodium ion concentration

*water follows sodium

Decrease ADH: diluted urine and decrease in volume in body fluids

Increase ADH: concentrated urine

Atrial natriuretic peptide: decrease BP and blood volume

Important buffer systems

- Respiratory system
- Kidneys

normal pH of body fluids

- Arterial blood = pH 7.4
- Venous blood and IF fluid = pH 7.35
- ICF = pH 7.0
- Alkalosis or alkalemia = arterial blood pH > 7.45
- Acidosis or acidemia = arterial pH < 7.35
- Physiological acidosis = pH between 7.0 - 7.35

Hydrogen ion regulation

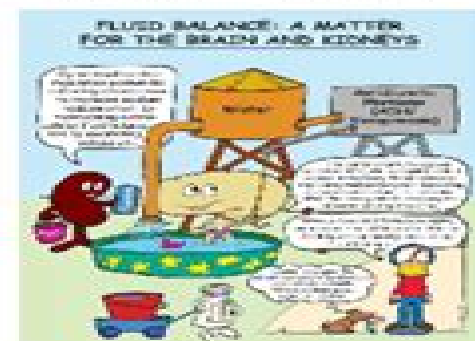
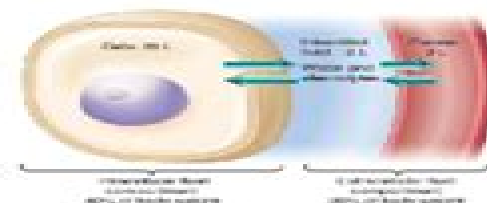
1. Chemical buffer system _____ (rapid) first line of defense
2. Respiratory centers - in brain stem act within 1 - 3 min
3. Renal mechanisms - _____ most potent

Three major chemical buffers

Chemical buffer: resist change in pH changes when a strong acid or base is added

1. Bicarbonate buffer system
2. Phosphate buffer system
3. Protein buffer system

Major fluid compartments in the body



Acid And Base Study Guide

Pamela Mills,Amina El-Ashmawy



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1. Understanding the eBook Acid And Base Study Guide
 - The Rise of Digital Reading Acid And Base Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Acid And Base Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acid And Base Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acid And Base Study Guide
 - Personalized Recommendations
 - Acid And Base Study Guide User Reviews and Ratings

- Acid And Base Study Guide and Bestseller Lists
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 - Highlighting and Note-Taking Acid And Base Study Guide
 - Interactive Elements Acid And Base Study Guide
- 8. Staying Engaged with Acid And Base Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acid And Base Study Guide
- 9. Balancing eBooks and Physical Books Acid And Base Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acid And Base Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Acid And Base Study Guide
 - Setting Reading Goals Acid And Base Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Acid And Base Study Guide
 - Fact-Checking eBook Content of Acid And Base Study Guide
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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