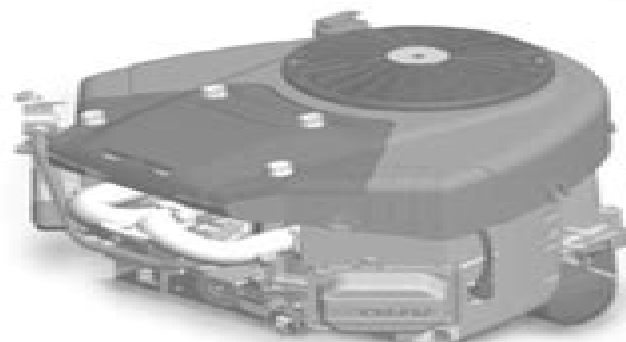
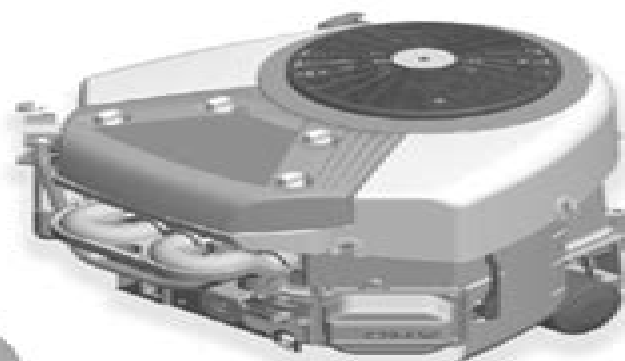




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hscn present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species
 $\text{mol Fe} = 3.170 \times 10^{-5} \text{ mol}$
 $\text{mol Fe} = 3.850 \times 10^{-4} \text{ mol}$

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