

# RADIOACTIVE DECAY

Radioactive Decay of Candium 150



**Introduction:** Some isotopes spontaneously emit particles or energy to reach a different, more stable, low-energy radioactive state. This decay occurs naturally and is not affected by chemical change, temperature, or pressure.

Radioactive decay takes place in a random way. Although you cannot predict just one given atom will decay, you can predict that all billions of atoms will in small pieces of a radioactive element, a given number will decay at a predictable and predictable rate.

Testing of radioactive elements in rocks leads to determine the absolute age of the rock. Comparing the percentage of an original element present today to the percentage of the element present today, the age of rock can be calculated. The ratio of the two atoms is a direct function of time because when the rock was formed, it must contain atoms of daughter atoms.

**Procedure:**

- Candium 150
- Half-life
- Radioactive decay
- Radioactive dating

4. Count and record the number of undecayed "undecayed" Candium atoms (M) and all remaining.

5. Record in the data table.

Repeat steps 2, 3 and 4 until all the radioactive "undecayed" (Parent M) atoms decay or 10 minutes in the case of radioactive decay.

| Half-life (M) | "Parent" atoms (M) of "undecayed" radioactive Candium atoms remaining after the "M" half-life | "Daughter" atoms (D) of "decayed" atoms |
|---------------|-----------------------------------------------------------------------------------------------|-----------------------------------------|
| 0             | 100                                                                                           | 0                                       |
| 1             |                                                                                               |                                         |
| 2             |                                                                                               |                                         |
| 3             |                                                                                               |                                         |
| 4             |                                                                                               |                                         |
| 5             |                                                                                               |                                         |
| 6             |                                                                                               |                                         |
| 7             |                                                                                               |                                         |
| 8             |                                                                                               |                                         |
| 9             |                                                                                               |                                         |
| 10            |                                                                                               |                                         |

**Graph:** Use the graph on the following page to represent the data collected throughout the procedure. The "parent" atoms (green) and "daughter" atoms (red) should be plotted on your graph.



## STUDENT LAB

# Activity 1 Radioactive Decay Lab Student Guide

**KJ Lindholm-Leary**



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