

CHAPTER 26 : ACID-BASE

Total body water volume: _____ this is 60% of body weight

Water: universal solvent

Solutes: nonelectrolytes and electrolytes

Extracellular and intracellular

- ECF : the fluid on the _____ the cell major cation is sodium and major anion is chloride
- Sodium is the most abundant cation in the extracellular fluid
- ICF: fluid _____ the cell major cation is potassium

Water intake is _____ to water output 2500 ml/day

Water output: urine, insensible water loss, perspiration, and feces

Water reabsorption: in collecting ducts proportional to ADH release

Aldosterone: important in regulation of sodium ion concentration

*water follows sodium

Decrease ADH: diluted urine and decrease in volume in body fluids

Increase ADH: concentrated urine

Atrial natriuretic peptide: decrease BP and blood volume

Important buffer systems

- Respiratory system
- Kidneys

normal pH of body fluids

- Arterial blood = pH 7.4
- Venous blood and IF fluid = pH 7.35
- ICF = pH 7.0
- Alkalosis or alkalemia = arterial blood pH > 7.45
- Acidosis or acidemia = arterial pH < 7.35
- Physiological acidosis = pH between 7.0 - 7.35

Hydrogen ion regulation

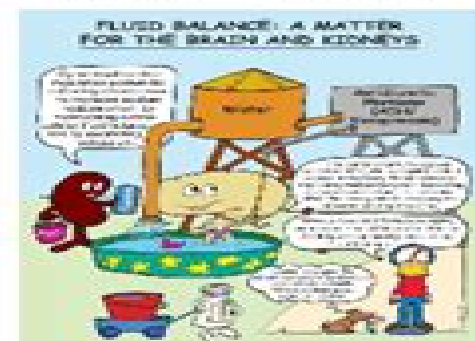
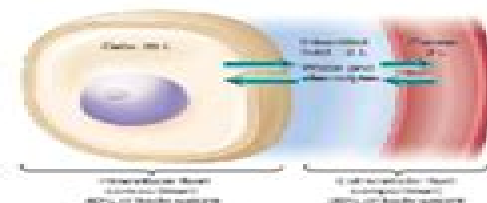
1. Chemical buffer system _____ (rapid) first line of defense
2. Respiratory centers - in brain stem act within 1 - 3 min
3. Renal mechanisms - _____ most potent

Three major chemical buffers

Chemical buffer: resist change in pH changes when a strong acid or base is added

1. Bicarbonate buffer system
2. Phosphate buffer system
3. Protein buffer system

Major fluid compartments in the body



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Gerard J. Criner, Gilbert E. D'Alonzo



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