

AP Chemistry: Acid-Base Worksheet

AP Chemistry Acid-Base WS 0809.doc

Names: _____ Date: _____ Per: _____

1. In the following equations, label each species as an acid or a base. Show the conjugate acid-base pairs.



2. In the following reactions, identify the Lewis acid and the Lewis base.



3. For the following reaction, decide which species (reactants or products) are favored at the completion of the reaction.



4. Which member of each of the following pairs is the stronger acid?



5. Calculate the concentrations of hydronium ion and hydroxide ion at 25°C in



6. A sample of orange juice has a hydronium-ion concentration of $2.9 \times 10^{-4}\text{M}$. What is the pH? Is the solution acidic?

7. The pH of human arterial blood is 7.40. What is the hydronium ion concentration?

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