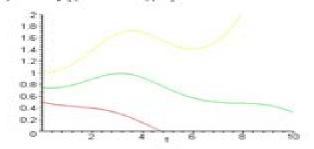
sides yields the general solution $y = [-k \int \mu(\tau) d\tau + y_0 \mu(0)]/\mu(t)$. In this problem, the integrating factor is $\mu(t) = \exp[(\cos t - t)/5]$.



(b). The population becomes extinct, if y(t*) = 0, for some t = t*. Referring to part(a), we find that y(t*) = 0 ⇒

$$\int_{0}^{t^{*}} exp[(\cos \tau - \tau)/5]d\tau = 5 e^{1/5}y_{c}.$$

It can be shown that the integral on the left hand side increases monotonically, from zero to a limiting value of approximately 5.0893. Hence extinction can happen only if $5\,e^{1/3}y_c < 5.0893$, that is, $y_c < 0.8333$.

(c). Repeating the argument in part(b), it follows that y(t*) = 0 ⇒

$$\int_{0}^{t^{*}} exp[(\cos \tau - \tau)/5]d\tau = \frac{1}{k} e^{1/5}y_{c}.$$

Hence extinction can happen only if $e^{1/5}y_c/k < 5.0893$, that is, $y_c < 4.1667 k$.

(d). Evidently, y_c is a linear function of the parameter k.

19(a). Let Q(t) be the volume of carbon monoxide in the room. The rate of increase of CO is $(.04)(0.1) = 0.004 \ ft^3/min$. The amount of CO leaves the room at a rate of $(0.1)Q(t)/1200 = Q(t)/12000 \ ft^3/min$. Hence the total rate of change is given by the differential equation dQ/dt = 0.004 - Q(t)/12000. This equation is linear and separable, with solution $Q(t) = 48 - 48 \exp(-t/12000) \ ft^3$. Note that $Q_0 = 0 \ ft^3$. Hence the concentration at any time is given by x(t) = Q(t)/1200 = Q(t)/12%.

(b). The concentration of CO in the room is x(t)=4-4exp(-t/12000) %. A level of 0.00012 corresponds to 0.012 %. Setting $x(\tau)=0.012$, the solution of the equation 4-4exp(-t/12000)=0.012 is $\tau\approx 36$ minutes.

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RD Boyd

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