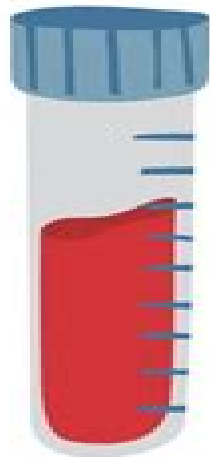
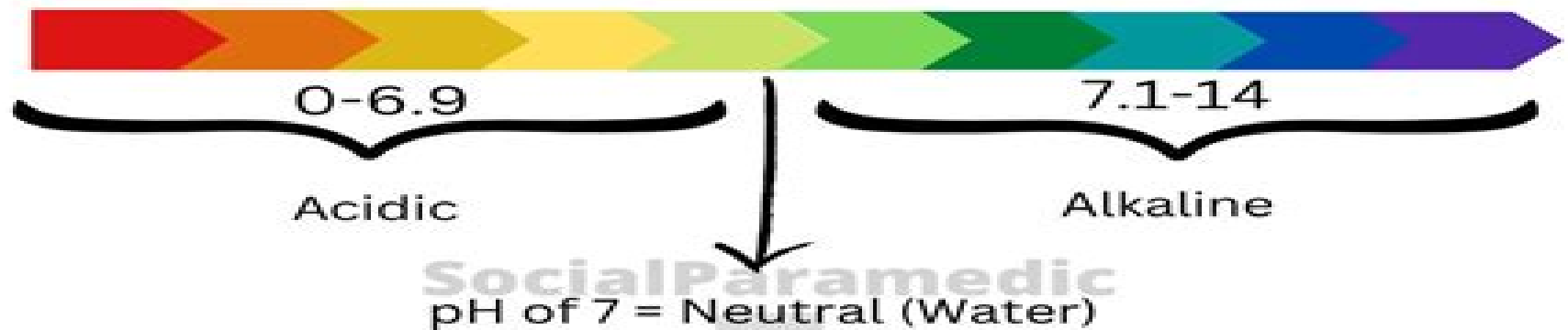


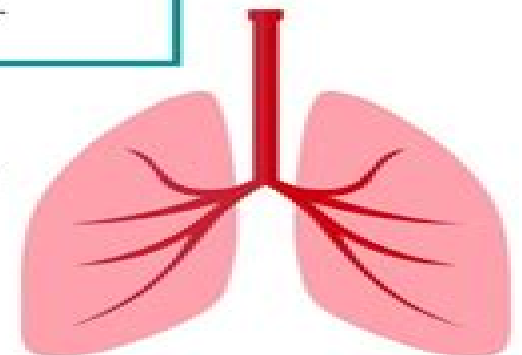
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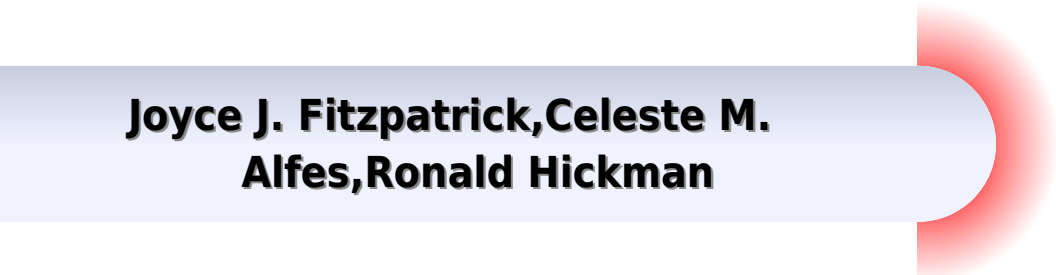
pH of Blood	7.35-7.45
ETCO ₂	35-45mmHg
HCO ₃ (Bicarbonate)	22-26 meq/L

HCO₃: Normally, O₂ combines with glucose to create energy and CO₂ is an acid waste by product of this. Bicarbonate is a buffer produced by the body that is alkaline, and binds with CO₂ to neutralize it to allow the body to get rid of it.



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**Joyce J. Fitzpatrick, Celeste M.
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