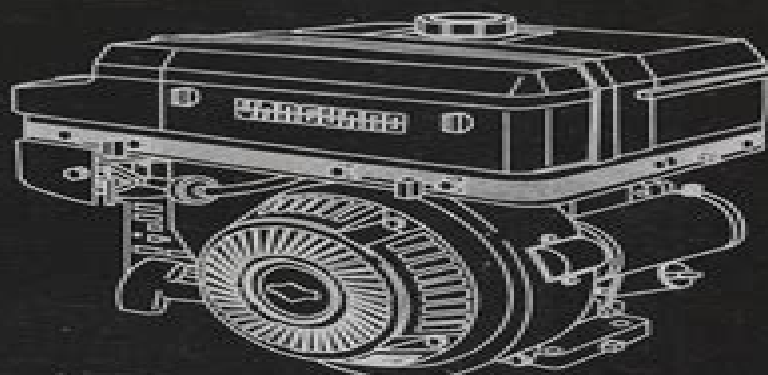




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web a iodine and sulphur respectively b iodine and hydrogen sulphide respectively c sulphur and iodine respectively d hydrogen sulphide and sulphur solve these important questions for class 10 science chapter 1 chemical reactions and equations before the exam to develop problem solving abilities prepare effectively for the exams

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web correct answer option b q3 the image shows some chemical reactions which option identifies the reactants and products of the reactions correct answer option c q4 a student performs an experiment to form aluminium chloride from aluminium and chlorine

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because after its combustion a new substance is formed and cannot be turned back into

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web q5 identify the reducing agent in the following reactions a $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$ b $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ c $\text{Fe} + 2\text{O}_3 \rightarrow \text{Fe}_2\text{O}_3 + 3\text{CO}_2$ d $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ answer a here ammonia NH_3 is the reducing agent b here water H_2O is the reducing agent