

Basic Stoichiometry Post-Lab Homework Exercises

1. Load the "Reactants, Products, and Leftovers" simulation and work through each of the levels of the Game! At <http://phet.colorado.edu/> or googling "phet." You may have to download or update Java on your computer.

Show your work for all calculations. Remember: Labels, units, sig figs and cancellation!

2. For the reaction $P_4 + 6Cl_2 \rightarrow 4PCl_3$, determine how many moles of chlorine Cl_2 would be needed to react with 4.0 moles of phosphorus P_4 to entirely use up all the phosphorus. 4)_____

3. If 3.00 moles of P_4 reacted with 22.0 moles Cl_2 according to the above reaction, determine:

- a) How many moles PCl_3 are produced a)_____
- b) How many moles of P_4 are left in excess after the reaction (if any) b)_____
- c) How many moles of Cl_2 are left in excess after the reaction (if any) c)_____

In reality, reactants don't have to react in perfect whole-numbers of moles. In a two-reactant synthesis reaction, usually **one reactant gets entirely used up (and determines how much product is made)**, even if that means using fractions of a mole of reactant. For instance, when solid, metallic aluminum Al and red, liquid bromine Br_2 are brought together, they make a white solid according to the reaction $2Al + 3Br_2 \rightarrow 2AlBr_3$. If 5.0 moles of aluminum Al was reacted with 10 moles bromine Br_2 , all five moles of aluminum would react, with only 7.5 moles bromine. (2:3 mole ratio) This would produce only 5.0 moles of $AlBr_3$, leaving 2.5 moles of excess Br_2 behind.

5. Now assume 3.0 moles Al and 4.0 moles Br_2 react.

- a) How much (in moles) $AlBr_3$ gets produced? a)_____
- b) Which chemical must be the *excess reactant*? b)_____
- c) Which chemical is the *limiting reactant*? c)_____
- d) If all the limiting reactant gets used up, how much of the excess reactant is left? d)_____

Basic Stoichiometry Phet Lab Homework Exercises Answers

**Frank Schaffer Publications,Joan
DiStasio**



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The book delves into Basic Stoichiometry Phet Lab Homework Exercises Answers. Basic Stoichiometry Phet Lab Homework Exercises Answers is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Basic Stoichiometry Phet Lab Homework Exercises Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Basic Stoichiometry Phet Lab Homework Exercises Answers
 - Chapter 2: Essential Elements of Basic Stoichiometry Phet Lab Homework Exercises Answers
 - Chapter 3: Basic Stoichiometry Phet Lab Homework Exercises Answers in Everyday Life
 - Chapter 4: Basic Stoichiometry Phet Lab Homework Exercises Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Basic Stoichiometry Phet Lab Homework Exercises Answers. The first chapter will explore what Basic Stoichiometry Phet Lab Homework Exercises Answers is, why Basic Stoichiometry Phet Lab Homework Exercises Answers is vital, and how to effectively learn about Basic Stoichiometry Phet Lab Homework Exercises Answers.
 3. In chapter 2, this book will delve into the foundational concepts of Basic Stoichiometry Phet Lab Homework Exercises Answers. This chapter will elucidate the essential principles that must be understood to grasp Basic Stoichiometry Phet Lab Homework Exercises Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of Basic Stoichiometry Phet Lab Homework Exercises Answers in daily life. This chapter will showcase real-world examples of how Basic Stoichiometry Phet Lab Homework Exercises Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Basic Stoichiometry Phet Lab Homework Exercises Answers in specific contexts. This chapter will explore how Basic Stoichiometry Phet Lab Homework Exercises Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Basic Stoichiometry Phet Lab Homework Exercises Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Basic Stoichiometry Phet Lab Homework Exercises Answers.

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