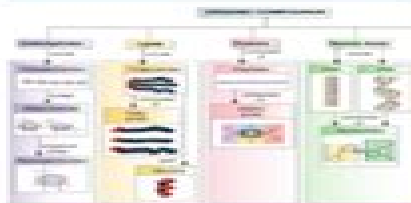


MACROMOLECULES

Common Functional Groups

Hydroxyl	-OH	polar
Carboxyl	-COOH	polar
Carbonyl	-CO	polar
Amino	-NH ₂	polar
Phosphate	-PO ₃	polar
Methyl	-CH ₃	non-polar
Sulfhydryl	-SH	non-polar

Organic Compounds Overview



*more details underneath this block

Carbohydrates

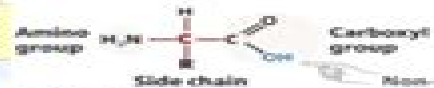
Monosaccharides	Disaccharides	Polysaccharides
- glucose	- sucrose	- "starch"
- fructose	- lactose	- cellulose
	- dextrose	- chitin
	- maltose	- amylose
	- galactose	- glycogen

*sugars can dissolve in H₂O because they are NON-POLAR!

Lipids

Triglycerides	Phospholipids	Steroids
-made of glycerol and fatty acids	-forms a bilayer (membranes)	-4 rings
-three "strands"	-polar, hydrophilic heads	-cholesterol
-no kink = saturated	-nonpolar, hydrophobic tails	-estrogen
-kink = unsaturated		-testosterone

Non-ionized form of amino acid



*R groups can be a mix of the following: unchanged, non-polar, polar, hydrophilic, hydrophobic

*amino acids are joined by a **PEPTIDE BOND**
*this results in a dehydration (taking H₂O out)

*protein = shape

Nucleotides

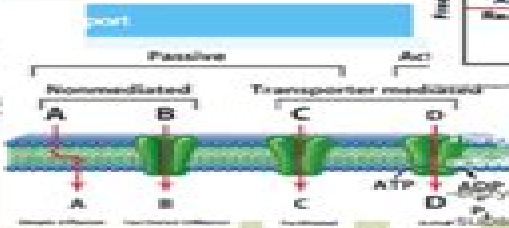


*amount of P_i (phosphate group) can change

TRANSPORT/ENZYMES

Osmosis

Hypertonic	Hypotonic	Isotonic
high solute, less water	low solute, high water	equilibrium
cell shriveled	cell swells (hippo)	cell is normal

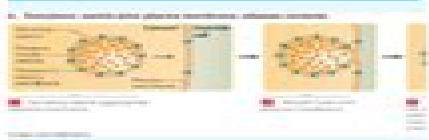


*more descriptions below

Endocytosis



Exocytosis



Enzymes

-bind to a substrate. These fit depending on the **SHAPE** of enzymes

-competitive inhibitors can block substrates by changing the shape of the enzyme

Enzymes are catalysts



*that factor enzyme

enzyme concentration
substrate concentration

-temperature
-pH

*enzymes have an optimal temp. and pH
*this depends on the specific enzyme!

*enzymes usually can go back to its original form, unless it **DENATURES**



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**Peter J. Mikulecky, Michelle Rose
Gilman, Brian Peterson**



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Table of Contents Biology 11 Final Exam Notes Cheat Sheet

1. Understanding the eBook Biology 11 Final Exam Notes Cheat Sheet
 - The Rise of Digital Reading Biology 11 Final Exam Notes Cheat Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology 11 Final Exam Notes Cheat Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology 11 Final Exam Notes Cheat Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biology 11 Final Exam Notes Cheat Sheet
 - Personalized Recommendations
 - Biology 11 Final Exam Notes Cheat Sheet User Reviews and Ratings
 - Biology 11 Final Exam Notes Cheat Sheet and Bestseller Lists
5. Accessing Biology 11 Final Exam Notes Cheat Sheet Free and Paid eBooks
 - Biology 11 Final Exam Notes Cheat Sheet Public Domain eBooks
 - Biology 11 Final Exam Notes Cheat Sheet eBook Subscription Services
 - Biology 11 Final Exam Notes Cheat Sheet Budget-Friendly Options
6. Navigating Biology 11 Final Exam Notes Cheat Sheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Biology 11 Final Exam Notes Cheat Sheet Compatibility with Devices
 - Biology 11 Final Exam Notes Cheat Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biology 11 Final Exam Notes Cheat Sheet
 - Highlighting and Note-Taking Biology 11 Final Exam Notes Cheat Sheet
 - Interactive Elements Biology 11 Final Exam Notes Cheat Sheet

8. Staying Engaged with Biology 11 Final Exam Notes Cheat Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology 11 Final Exam Notes Cheat Sheet
9. Balancing eBooks and Physical Books Biology 11 Final Exam Notes Cheat Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology 11 Final Exam Notes Cheat Sheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Biology 11 Final Exam Notes Cheat Sheet
 - Setting Reading Goals Biology 11 Final Exam Notes Cheat Sheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Biology 11 Final Exam Notes Cheat Sheet
 - Fact-Checking eBook Content of Biology 11 Final Exam Notes Cheat Sheet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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