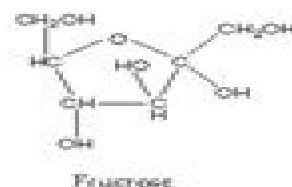
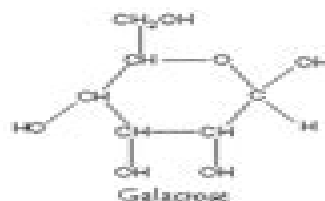
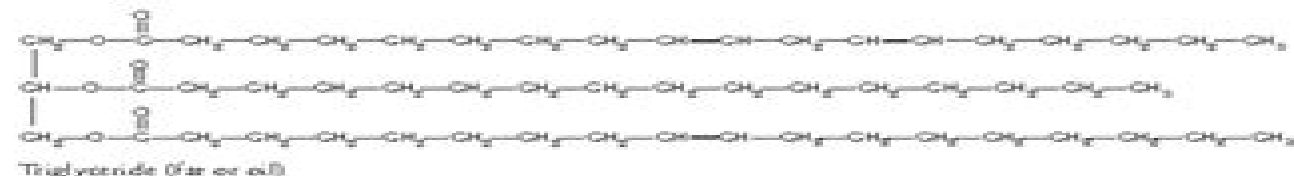


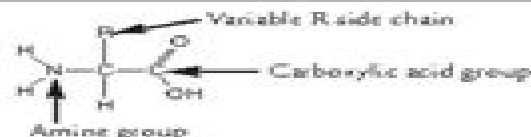
What are the building blocks of life?

From the smallest single-celled organism to the tallest tree, all life depends on the properties and reactions of four classes of **organic** (carbon-based) compounds—**carbohydrates**, **lipids**, **proteins**, and **nucleic acids**. These organic molecules are the building blocks of all living things, and are responsible for most of the structure and functions of the body, including energy storage, insulation, growth, repair, communication, and transfer of hereditary information. Simple organic molecules can be joined together to form all the essential biological molecules needed for life.

Carbohydrates (monosaccharides)


$$\begin{array}{c} \text{H} \\ | \\ \text{H}-\text{C}-\text{OH} \\ | \\ \text{H}-\text{C}-\text{OH} \\ | \\ \text{H}-\text{C}-\text{OH} \\ | \\ \text{H} \end{array}$$
C[C@@H](C(=O)O)N

Alanine



The diagram illustrates the chemical structure of a nucleotide, the basic building block of nucleic acids. It consists of three main components: a phosphate group, a sugar, and a nitrogenous base. The phosphate group is shown on the left, consisting of a phosphorus atom (P) bonded to four oxygen atoms (O), one of which is part of a hydroxyl group (OH). This phosphate group is linked to a five-carbon sugar molecule (deoxyribose) via a phosphodiester bond. The sugar is a pentose ring with hydroxyl groups at the 2' and 3' positions. Attached to the 1' carbon of the sugar is a nitrogenous base, which is a purine derivative (adenine) with an amino group (NH₂) at the 6-position. Arrows point from the labels 'Phosphate group', 'Sugar', and 'Nitrogenous base' to their respective parts in the structure.

Biological Molecules Pogil

LL Leslie



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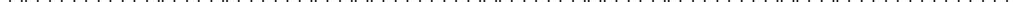
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