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reactions $6e^-$ $14H^+$ $Cr_2O_7^{2-}$ $2Cr^{3+}$ $7H_2O$ Fe^{2+} Fe^{3+} e^- 2 equalize the electrons $6e^-$ $14H^+$ $Cr_2O_7^{2-}$ $2Cr^{3+}$ $7H_2O$ $6Fe^{2+}$ $6Fe^{3+}$ $6e^-$ multiplied by a factor of 6

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