

INSTALLATION INSTRUCTIONS CARE AND MAINTENANCE

American Standard

Style That Works Better

FloWise™ DUAL FLUSH 2-PIECE TOILET

Model 2479.216 / 2479.516 / 2476.216 / 2476.516

Model 2480.216 / 2480.516 / 2484.216 / 2484.516 Right Height™



Thank you for selecting American Standard - the benchmark of fine quality for over 100 years. To ensure this product is installed properly, please read these instructions carefully before you begin. (Certain installations may require professional help.) Also be sure your installation conforms to local codes.

CAUTION: PRODUCT IS FRAGILE. TO AVOID BREAKAGE AND POSSIBLE INJURY HANDLE WITH CARE!

NOTE: Pictures may not exactly define contour of china and components.

RECOMMENDED TOOLS AND MATERIALS

Putty Knife
Hack Saw

Regular Screwdriver
Wax Ring/Gasket

Adjustable Wrench
Flexible Supply Tube

Sealant
Closet Bolts

Tape Measure
Carpenter's Level

1 REMOVE OLD TOILET

- Close toilet supply valve and flush tank completely. Towel or sponge remaining water from tank and bowl.
- Disconnect and remove supply line.

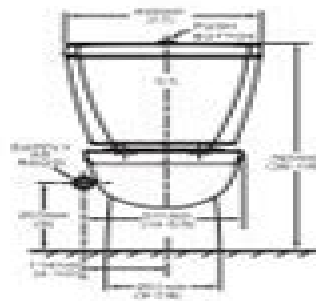
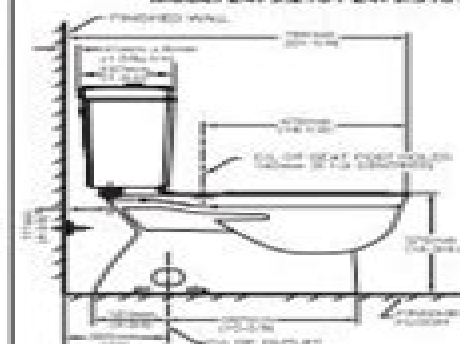
NOTE: If replacing valve, first shut off main water supply!

- Remove old mounting hardware, remove toilet and plug floor waste opening to prevent escaping sewer gases.
- Remove closet bolts from flange and clean away old wax, putty, etc. from base area.

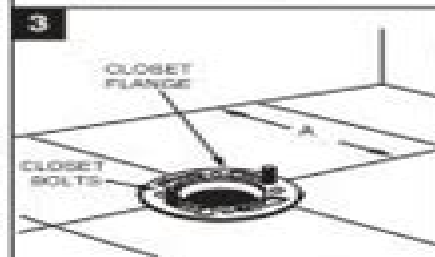
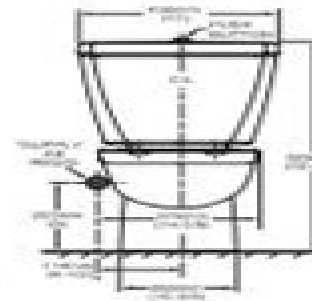
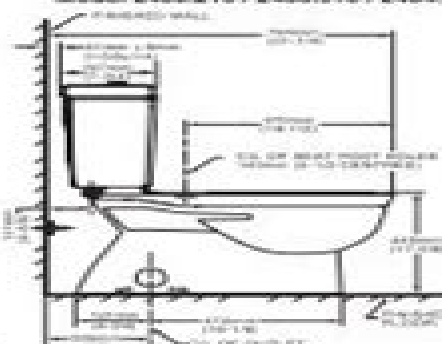
NOTE: Mounting surface must be clean and level **before** new toilet is installed!

2 ROUGHING-IN DIMENSIONS:

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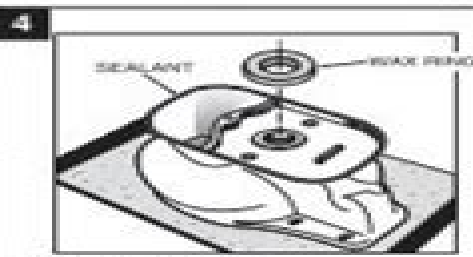


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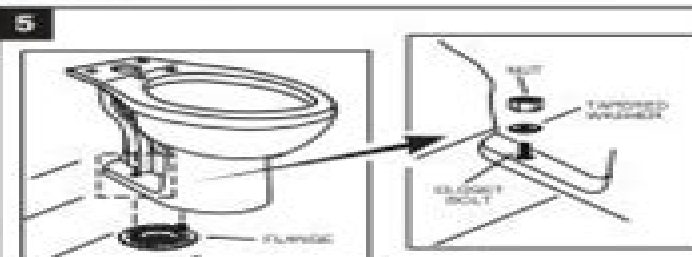
3 INSTALL CLOSET BOLTS

Install closet bolts in flange channel, turn 90°, and slide into place 6" (152mm) apart and parallel to wall.



4 INSTALL WAX SEAL

Insert toilet on floor (ouchion to prevent damage), and install wax ring evenly around waste flange (horn), with tapered end of ring facing toilet. Apply a thin bead of sealant around toilet base.



5 POSITION TOILET ON FLANGE

- Unplug floor waste opening and install toilet on closet flange so bolts project through mounting holes.
- Loosely install retainer washers and nuts. Side of washers marked "THIS SIDE UP" must face up!

SAVE FOR FUTURE USE

American Standard Acont402 Installation Manual

Mark Steyvers



American Standard Acont402 Installation Manual:

90.1 User's Manual American Society of Heating, Refrigerating and Air-Conditioning Engineers, 2004 This User's Manual provides detailed instruction for the design of commercial and high rise residential buildings to ensure their compliance with ANSI ASHRAE IESNA Standard 90.1 2004 In addition this Manual encourages the user to apply the principles of effective energy conserving design when designing buildings and building systems offers information on the intent and application of Standard 90.1 illuminates the Standard through the use of abundant sample calculations and examples streamlines the process of showing compliance provides Standard forms to demonstrate compliance provides useful reference material to assist designers in efficiently completing a successful and complying design This Manual also instructs the user in the application of several tools used for compliance with Standard 90.1 the EnvStd computer program used in conjunction with the Building Envelope Trade Off compliance method the selection and application of energy simulation programs used in conjunction with the energy cost budget method of compliance This Manual is intended to be useful to numerous types of building professionals including architects and engineers who must apply the Standard to the design of their buildings plan examiners and field inspectors who must enforce the Standard in areas where it is adopted as code general and specialty contractors who must construct buildings in compliance with the standard product manufacturers state and local energy offices policy groups utilities and others **Manual of Acceptable Practices [for] ASHRAE Standard 90**, 1980 SA98 Installation Manual Sophia Systems, 1988 *90.1 User's Manual ANSI/ASHRAE/IESNA Standard 90.1 - 2001* American Society of Heating, Refrigerating and Air-Conditioning Engineers, 2002 The 90.1 User's Manual was developed as a companion document to ASHRAE IESNA Standard 90.1 2001 and reflects all addenda and changes made to the standard The User's Manual eases use of the standard by offering information about its intent and application as well as by including numerous examples and sample calculations that illustrate how architects and engineers can apply Standard 90.1 2001 to their building designs The manual streamlines the compliance process and includes standard ready to use compliance forms It also provides information on energy simulation computer programs used in the energy cost budget method of compliance A CD accompanies the manual and contains an updated version of the EnvStd computer program and PDF versions of the compliance forms provided in the User's Manual The EnvStd program is used for doing building envelope trade offs The CD requires a 486 or Pentium based computer and either Microsoft Windows 95 or Windows NT 3.5 or later 8MB of RAM 16MB recommended and 10MB of free hard disk space is required User's Manual for ANSI/ASHRAE Standard 15 2001, Safety Standard for Refrigeration Systems D. L. Fenton, William V. Richards, 2003 This user's manual was developed as a companion document to ASHRAE Standard 15 2001 It does not reflect the addenda and changes incorporated into Standard 15 2004 The User's Manual clarifies the intent of the Standard and provides an explanation of the rationale behind it It eases use of the standard by including illustrations and examples of accepted industry practice as well as explanations of and supporting

references for formulas in the Standard This guide also covers building system and refrigerant classifications restrictions on refrigerant use installation restrictions and equipment and system design and construction The User s Manual includes information on mechanical and absorption refrigeration systems for commercial residential and industrial applications

90.1 User's Manual American Society of Heating, Refrigerating and Air-Conditioning Engineers,Illuminating Engineering Society of North America,American National Standards Institute,2007 HP 30018A/30019A Add-On ADCC Installation Manual Hewlett-Packard Company,1981 *90.1 User's Manual - ASHRAE Standard 90.1-2007 User's Manual* ,2007 Manual of Acceptable Practices for ASHRAE Standard 90, 1975 ,1980-01-01 ANSI/ASHRAE Standard 62.1 User's Manual ,2010

Reviewing **American Standard Acont402 Installation Manual**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**American Standard Acont402 Installation Manual**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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web google classroom for the following redox equation $\text{MnO}_4^- + \text{H}^+ + \text{SO}_2 \rightarrow \text{Mn}^{2+} + \text{H}_2\text{O} + \text{HSO}_4^-$ what will be the stoichiometric coefficient of HSO_4^- if the above equation is balanced in an acidic medium stuck use a hint report a problem do 4 problems

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web balance this redox reaction by the half reaction method by assuming an acidic solution $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} \rightarrow \text{Cr}^{3+} + \text{Fe}^{3+}$ balance the redox reaction in exercise 9 by the half reaction method by assuming a basic solution

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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^- + \text{Au} \rightarrow \text{AuCN}_2^- + 2\text{e}^-$ $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ make electrons equal convert to basic solution $4\text{CN}^- + 2\text{Au} \rightarrow 2\text{AuCN}_2^- + 2\text{e}^-$ multiplied by a factor of 2 $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ the final answer

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web aug 29 2023 the first step to balance any redox reaction is to separate the reaction into half reactions the substance being reduced will have electrons as reactants and the oxidized substance will have electrons as products usually all reactions are written as reduction reactions in half reaction tables

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balancing redox reactions in acidic solution problems 1 10 problem 1 $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} \rightarrow \text{Cr}^{3+} + \text{Fe}^{3+}$ solution 1 balanced half reactions $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} + \text{e}^-$ 2 equalize the electrons $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $6\text{Fe}^{2+} \rightarrow 6\text{Fe}^{3+} + 6\text{e}^-$ multiplied by a factor of 6

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web when balancing equations for redox reactions occurring in acidic solution it is often necessary to add H^+ ions or the H_2O pair to fully balance the equation in this video we ll walk through this process for the reaction between dichromate $\text{Cr}_2\text{O}_7^{2-}$ and chloride Cl^- ions in acidic solution created by jay

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web balance each half reaction in basic solution $4\text{Fe}^{2+} + \text{O}_2 + 10\text{OH}^- \rightarrow 4\text{Fe}^{3+} + 2\text{H}_2\text{O} + 8\text{OH}^-$ $\text{Mn}^{2+} + \text{H}_2\text{O} + 2\text{OH}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$ balance each redox reaction in acid solution using the half reaction method $8\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} + 2\text{e}^- \rightarrow 2\text{Cr}^{3+} + 4\text{H}_2\text{O}$ $2\text{H}_2\text{O} + \text{N}_2\text{O}_4 \rightarrow \text{O}_2 + 2\text{H}^+ + 2\text{H}_2\text{O}$ $\text{Cr}^{3+} + \text{Te}^{10+} \rightarrow \text{ReO}_4^- + \text{PbO}_2 + \text{IO}_2 + \text{Pb}^{12+}$ as $\text{NO}_3^- + \text{IO}_3^- + \text{H}_2\text{AsO}_4^-$

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